

arts dates back at least to the ancient Egyptians. One of its first uses was in glass-making and analyses of Egyptian and Roman glassware have shown the presence of two per cent manganese.

Pliny, in his treatise on natural history, mentions the use of manganese among the Romans in decolorizing glass. He considered it a variety of magnetic iron ore, or lodestone, which he speaks of as magnes. During the middle ages it was known as magnesia, and was later referred to as magnesia alba.

The ancient, however, overlooked a few

things in connection with manganese in the manufacture of glass. Modern scientists have about concluded that it is the small per cent of manganese in nearly all the glass sand from which glass is manufactured, that keeps the ultra violet ray of the sun from passing through common window glass. We might add that north Arkansas has glass sand deposits which contain no manganese, and probably in the more modern days to come will be very valuable for this reason.

Col. Matt Martin of Batesville was the first to discover the value of manganese

ore in what is now the Batesville-Cushman manganese field. Between 1848 and 1850, he and M. D. Fields acquired large tracts of land in the ore-bearing region. Colonel Martin worked this land to a limited extent and as early as 1850 shipped small quantities of the ore to Boston, New York and Philadelphia. The ore shipped during this period was used for chemical purposes and not for the manufacture of steel, for which it is most in demand today. Mining operations have been carried on in the field almost constantly. The production has never been very large, but it has been steady, and affords work for several hundred men during normal times.

Nature is always playing tricks on the mining profession. In the Batesville-Cushman field until last year, all the ore mined was boulder or fine ore, found in clay. It was mined from shafts, drifts and open cuts. Some of the ore pockets were large, some small, but there were enough large ones to make it a paying industry. The ore they produced is an oxide ore with a blue cast.

About a year ago a miner sent in a load of this ore in the form of round, smooth boulders. They looked suspicious to Mr. Baker, who had charge of the buying, and he broke one of the boulders with a sledge hammer. The blue ore was only a veneer on what appeared to be a core of gray rock, or limestone, and he turned the load down. Later this gray rock core was assayed and found to be rich manganese ore.

Investigations proved that this was carbonate of manganese, and blanket veins of this ore are found in many parts of the district, which has increased the reserves of the district. This investigation also showed that all of the oxide ore was once carbonate, and that the carbonates were the original ore bodies of the field. As erosion took place, the carbonate veins were broken down, and on coming into contact with the elements changed their chemical nature from a carbonate to an oxide.

Company Installs Machinery To Work Manganese Mine.

Special to the Gazette, June 21, 34
Cushman, June 20.—Walter H. Denison, and his sons, Reed and Alvis, are installing drilling machinery on a new manganese ore strike made on the Club House property a mile west of here. The equipment will consist of three engines and compressors and three jack hammer drills. The old Club House manganese mine has been in operation for the last 25 years and was a large producer before and during the World war.

The new strike was made on the same property across a hollow from the old digging. The vein has been opened with a 30-foot breast. It varies in thickness from about 18 inches to five or six feet. The ore is ledge ore, consisting of both carbonates and oxides. Alvis Denison, mining engineer, has charge of the new operations. A laboratory for assaying purposes has been installed in the company's office here.

MANGANESE MINING GREATLY INCREASED

Farmers of North Arkansas Regaining Losses From Drouth.

Special to the Gazette, Aug. 3, 1934
Cushman, Aug. 2.—More manganese ore has been mined and shipped from the Batesville-Cushman field during the last 30 days than was shipped during the entire year of 1933. Reed Denison, ore buyer, reported shipment of 1,300 tons, most of which went to the furnaces at Birmingham. Farmers who lost their crops by the drouth are regaining their losses by mining. An executive order signed by the president in March, to the effect that all material entering in government contracts had to be of domestic origin, is helping the demand for ore in this field. Tom Shell and A. B. Reither have 35 men working at the Martin mine near here, which they recently opened. They are taking a good production from a four-foot vein of carbonate ore.

TARIFF NEEDED BY MANGANESE MINERS

Arkansas Operators Protest Removal or Reduction of Duty.

(From the Gazette's Correspondent.)
Washington, Aug. 6.—When America entered the World war it possessed within its own boundaries sufficient quantities of all but seven of the minerals necessary for munitions and war material. One of the seven it lacked was manganese, a metallic element found principally in Montana, Washington and Arkansas, which has taken its place as an essential in an alloy for steel and copper. Mixed with these metals it increases their hardness, tenacity, and elasticity in the gears and wearing parts of heavy machinery, ordnance, and armor plate.

On August 3, a meeting of manganese producers, operating small throughout the country, was held in Washington with a view to forestalling attempts of consumers to have protective tariff removed or reduced and also with a view to furthering research work of governmental bureaus looking to more practical mining practices and improvements in beneficiating processes whereby the standard of low grade ores is raised.

So important was this meeting considered to Arkansas that Acting Governor Parnell dispatched George C. Branner, state geologist, to Washington to attend the proceedings and report developments.

In view of this interest in a local product which may some day prove of considerable value to the state it may be well to describe some of the characteristics of this little known and elusive mineral.

Resembles Iron.
Manganese resembles iron but is not magnetic. It has a grayish luster faintly tinged with red, and rusts rapidly in moist air. Pure manganese is only a chemical curiosity. "Speigeleisen" is an iron with about 12 per cent manganese, and is made in a special high-temperature furnace. "ferro-manganese" has about 70 per cent. Both are largely used as steel alloys and are added to the molten steel as it comes from the converter or open hearth furnace. Salts derived from manganese are used in chemistry, medicine and the arts. "Condy's fluid," a familiar disinfectant, is sodium permanganate in solution.

Prior to 1914, Russia was the greatest producer of high grade manganese. Brazil and India ranked next. The use of the low-grade deposits in the United States involves too heavy expense, under present mining methods, to be very profitable in normal times.

One day in 1868 an English metallurgist, Robert F. Mushet, noticed that a steel tool with which he was cutting metal at a lathe did not lose its edge even when heated red hot by friction. He had the steel analyzed and found that it was an alloy or mixture of steel and tungsten, a metal little known in those days.

Since then it has been found that certain other metals, when mixed with steel in just the right proportions, will give the alloy this same quality. Among them are chromium, vanadium, cobalt, manganese and molybdenum.

Revolutionized Metal Working.
The discovery of these "high speed" steels, as they are called, revolutionized metal working the world over.

Although nearly all metals are used in alloy forms, the alloys of steel and iron are by far the most numerous and the most useful. Nor does the nature of an alloy depend alone on the kind of metal used, for often it is greatly altered by varying the proportions of the same metal.

It is possible for great flying machines to skim through the air because engines can be built of these new alloys that weigh no more than two pounds per horsepower. When a proportion of manganese is added to steel, it becomes so hard that it cannot be machined, but has to be cast or ground into shape. This steel makes armor plate that can not be pierced except with projectiles pointed with the same material.

A combination of nickel steel with manganese, silicon, and chromium gives the hard alloy armor, which is much used for measuring implements and pendulum rods because it is practically nonexpansive within normal temperature variations.

Manganese also is used in the manufacture of colored glassware, which is made by adding small amounts of various metallic oxides. Chromium give green shades, manganese shades from pink to purple; copper yields reds and blacks; uranium, yellows and greens, and cobalt a great variety of blues.

Tariff Encourages Operators.
The tariff on manganese continues to encourage new operators to become producers of manganese, and a large number of new producers have become active during the past three years. However, the number of shippers who contributed only one or two carloads to the total is surprisingly large. This condition is exemplified in Georgia, where there were eight shippers of high grade ore in 1924 who made total shipments of only 1,093 gross tons, an average of less than 140 tons for each operator.

The world at present depends upon four great sources for its manganese: Russia, India, Brazil and the Gold Coast. In 1925, the world's production was approximately 2,570,000 tons, of which Russia produced 32 per cent; India 24 per cent; Brazil 12 per cent and the Gold Coast 13 per cent. The Tchiaturi deposit, in the Province of Kutais, Republic of Georgia, south of the Caucasus, may be classed as the largest single deposit of manganese ores now known. It covers some 22 square miles, and estimates of the total commercial ore range from 44,000,000 to 200,000,000 tons.

Demand Exceeds Supply.
In the United States the production of manganese, except during a few years prior to the advent of the Bessemer and open-hearth processes for making steel, has never been able to satisfy the domestic demand. The occurrences of manganese in this country are, broadly speaking, well known, and if the past history of the mining of this metal is taken to indicate the trend of the future, the United States does not possess a supply that can meet its demands. Unless methods are evolved for beneficiating low-grade manganiferous ores, this country must remain dependent upon foreign sources as long as manganese is required in the manufacture of steel.

Mining of manganese in the United States began in 1832 at the Paddy Run Mine, Frederick county, Va., and during the first few years of its operation

some 5,000 tons of high-grade ore was shipped to England.

Since 1890 the domestic production has ranged from a fraction of one per cent of the domestic needs to some 32 per cent during the war years of 1917 and 1918. The ores produced during these years were mixed with high grade imported ores in order to render them usable. And their utilization was made possible only by modification of peace-time specifications. Except for the chemical ores produced in the Phil-

lipsburg district of Montana, the supply of high-grade metallurgical manganese ores in the United States is meager.

DEBATE HEARD ON PRODUCTION OF MANGANESE

Feb. 8, 1936

Steel Men Oppose Developing Industry

Washington, Feb. 7 (AP).—Prompt development of the nation's manganese resources to permit a quick war-time expansion of steel manufacturing was urged before the Senate Munitions Committee today.

Representatives of the steel industry agreed that without manganese there can be no steel. However they argued that such a development would exhaust America's readily available supply, and force the nation to depend on imports in case of war.

Behind the argument lay a tariff dispute long before Congress. Manganese producers have been seeking a stiff rate on imports. Steel companies have been asking a low tariff. A duty of 14 cents a pound, imposed by the tariff law of 1930, was cut in two by the reciprocal tariff treaty with Brazil.

J. Carson Adkerson, president of the American Manganese Producers Association, argued that the 1930 rate should be restored to encourage development of the mines and have them producing should an emergency arise.

He reminded the committee that the War Department has proposed that the government acquire a stock of 600,000 tons of manganese. He said that he was informed that the War Department recently has increased its estimate of emergency need to 1,000,000 pounds. He said that this could be supplied, for a cost of \$30,000,000.

He intimated the investment of the manganese industry at \$20,000,000. Adkerson also said that during the 1920s, the steel industry bought foreign manganese at prices \$4 a ton higher than the delivered price of similar manganese from domestic mines.

Steel Producers Argue.
W. H. Johnstone of Bethlehem Steel replied that his company relied upon imports because of "price, grade and assured quantities of delivered supply."

He said Bethlehem had no interest in any manganese deposit outside this country, and denied a report that Bethlehem was obligated to accept a specific portion of imports from Russia.

Profit Figures Given.
During the day three contracts taken at random by committee investigators from the files of the Carnegie Steel Company showed profits ranging from 42.7 per cent of cost on orders for steel armor plate supplied to the navy. The profit figures on the three orders were \$130,964, \$64,919 and \$43,418.

STEEL, MANGANESE, TARIFF AND ARKANSAS.

Feb 10, 1936
In the controversy provoked by the 50 per cent cut in manganese import duties provided for in Secretary Hull's reciprocal trade agreement with Brazil there is another example of the complex problems inherent in tariffs.

We can not sell abroad unless we buy from abroad. A tariff law that unduly restricts imports reacts on export opportunities. Secretary Hull has been seeking to re-open clogged trade channels by reciprocal concessions. But every time he concedes anything on the part of the United States the affected domestic interest protests.

In the case of Brazil and manganese, the great steel interests are for the concession. They undertake to justify their position by arguing that America should depend on foreign sources for its ordinary supply of manganese, indispensable for steel making, and conserve its own manganese ore lest the readily available supply be exhausted and the country have to depend on imports in a war emergency.

But the manganese interests retort that what national safety dictates is the development now of domestic sources, so that the country may be ready for a war emergency. Moreover, steel is firmly entrenched behind a high tariff wall, which makes its attitude as an opponent of protection

for other metallurgical industries somewhat awkward.

About half of the country's annual consumption of 500,000 tons of manganese ore represents domestic production. (Comparisons are difficult because of the varying manganese content of ores.) Arkansas's stake in manganese mining may seem to be comparatively small. In 1917 our production of high and low grade manganese ores, stimulated by World war demand and prices, rose to nearly 20,000 tons. At present it runs from 2,000 to 5,000 tons a year, all grades. But still Arkansas manganese producers might inquire how, in fairness, they can be asked to jeopardize their modest production while steel tariffs remain unchanged.

Manganese Field May Be Aided by Act

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Cushman, May 29.--The Batesville-Cushman manganese field expects to benefit from the naval appropriations bill passed by Congress recently and which carried a provision for the expenditure of \$3,500,000 for strategic war minerals. Manganese is probably the most important mineral on the strategic war mineral list, and Arkansas will probably contribute its quota to the purchase.

There are other minerals in the strategic minerals list, and just how much of the \$3,500,000 will be spent for manganese is yet undetermined. It will depend largely upon how the joint board representing the army and navy, in the purchases divides it. The plan before the bill was passed was to create a stock pile of 250,000 tons of ferromanganese, which is the metal obtained from manganese ore, which would have taken from 500,000 to 750,000 tons of ore to produce. The appropriation was slashed when the bill was passed, and the present appropriation will not permit any such tonnage to be purchased.

Congressman John E. Miller of the Second congressional district, who was instrumental in having the provision for the purchase of strategic minerals attached to the bill, said: "It was provided in the bill as originally passed in the House that only strategic minerals should be purchased, and they should be purchased from American producers. However, the compromise provision does not provide that only home produced minerals shall be purchased, but we have a law that provides that the army and navy shall purchase materials that are produced in America, whenever the material is available, and relying on the general law, we agreed to accept the compromise."

The idea of the government acquiring a large stock pile of ferromanganese was probably born from the experience of the government during the World war. When this country entered the conflict, most of the manganese ore entering into the manufacture of steel, was being imported from Russia and Brazil. Owing to the hazards of ocean transportation these imports were cut off, and the steel industry had to turn to domestic producers for this necessary war metal. For nearly two years government agencies urged every manganese mine operator to produce his maximum output, and encouraged miners of other ores to enter into the industry, as the nation was in dire need of it for munition purposes.

With a big reserve of manganese on hand, the situation would be different. If another was broke out domestic producers would have time to get mining operations under full headway before the stock pile was exhausted.

LONE MINER SHOWS PATIENCE OF A JOB Gazette 6-20-37 IN MANGANESE AREA

By JOHNNY ERP.

Batesville, June 19.—Job, the patient man of Biblical days, has a counterpart in S. S. ("Red") Gilbert, Cushman manganese miner.

Job was afflicted with ills and physical ailments.

"Red" was worried with an idea.

In fact, he's still bothered.

But like Job—he's patient.

"Red" has something else besides patience. That is faith.

And with this faith, patience, a good right arm, and a pick and shovel, "Red" set about to move a mountain—a shovel full at a time.

Heard Old Story.

"Red" got his idea back years ago when he heard the story how Henry Mallet, a miner, had sunk a shaft down on a rich lead of manganese ore in the grub-cut mining field, one mile north of Cushman.

Mallet, a well known miner of the the mining industry and the "lost lode" became just a myth and people began to doubt that it even existed.

That is—they all did—but "Red."

That was in 1932.

With all the confidence of a sophomore on freshman stunt night, "Red" picked up a pick and a shovel, a fresh chaw of "fishhook" and moved down to the bottom of the hill and started a six and one-half foot by four and a half foot tunnel in the general direction of the mythical vein.

"Red" used a wheelbarrow at first to haul out the dirt and refuse.

But after a year or so of this back-breaking exercise, he built a track and has since used a small dump car.

Time elapsed.

"Red" kept on digging.

Now, with nearly five years passed, during which "Red" has picked his way slowly through flint and soil, living half of his time from the sunlight, and enduring discomforts under ground in a mining tunnel, "Red" is still patient.

He has staked his life daily against the dangers of poison air pockets and body-crushing cave-ins. That hasn't worried him a bit.

Today, his tunnel reaches 200 yards back into the hill and is now only a few yards from the supposed location of the "lost lode."

"Red" may reach the end of his rainbow deep down under the surface of the earth. Any minute may bring about the fulfillment of what dubious-minded men call "a pipe dream." Any strike of the pick may reveal the blue, shining vein that has caused a man to transform himself into a human badger and for the last five years to meander around in the bowels of Mother Earth, converting a mountain into a veritable shell.

"Red" seems to the least worried man in these parts.

Like Job—he's patient!