

STATE OF ARKANSAS

ARKANSAS GEOLOGICAL COMMISSION
William V. Bush, State Geologist

WATER RESOURCES CIRCULAR NO. 17

Roubidoux Formation and Gunter Sandstone Member of the Gasconade Formation,
Major Aquifers in northern Arkansas

by

William L. Prior, J. Michael Howard, John David McFarland, and Steven S. Hill



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Abstract

The Roubidoux Formation and the Gunter Sandstone Member of the Gasconade Formation, both of Early Ordovician age, are important aquifers in the Ozark Plateau region of northern Arkansas. These units are composed of sandstone and dolostone. They are the preferred aquifers in this region of Arkansas because they yield relatively high volumes of potable water and are deep enough to avoid surface water infiltration.

The two units are exposed in Missouri, but are present only in the subsurface in northern Arkansas. Because they dip to the south at 0.5 to 2.0 degrees (46 to 184 feet per mile), their depth increases to 3,000 feet or more in the Boston Mountains area. South of the Boston Mountains, the Roubidoux Formation was encountered at a depth of greater than 19,000 feet in one well included in the study.

Well yields in the Roubidoux Formation and Gunter Sandstone Member are affected by the following factors: depth of penetration of the well bore into the water-producing unit, total thickness, relative proportion of sandstone to dolostone, degree of cementation of sandstone, and secondary porosity/permeability from recrystallization of the dolostone.

Chemically the water is of good quality—a magnesium-bicarbonate type with generally low chloride content. Some wells in the northwestern portion of the study area have relatively high chloride concentrations, but are still within accepted standards for potable water.

The impact on water quality and yield due to increased usage of potable water from these units has not been determined.

Introduction

The Roubidoux Formation and the Gunter Sandstone Member of the Gasconade Formation are used extensively as sources of ground water for the public, commerce, industry, and rural water districts. Recently, deeper wells to the Roubidoux were required in order to meet increased agricultural water needs. In properly completed wells, these units

receive no surface contamination because of their depth. Consequently, the Arkansas Department of Health has instructed many public water-supply utilities in northern Arkansas to convert from shallow wells and springs that are susceptible to contamination and utilize the Roubidoux and/or Gunter.

This report summarizes geologic and hydrologic data available for the

Roubidoux and Gunter Sandstone in northern Arkansas. It involved gathering and studying drill cuttings; preparing and interpreting both geologic and drilling logs to determine stratigraphic relations; and researching and tabulating water levels and yields reported by other workers. Well cuttings and strip logs were evaluated for a total of 71 wells. Finally, digital models, utilizing Surfer Surface Mapping System™ (version 6.02) by Golden Software, Inc., were built in which the data set was gridded by the Kriging method to allow extrapolation for contours drawn for geologic and hydrogeologic maps.

Several ground-water investigations of the Ozark Plateaus province have been conducted: Durfor and Becker (1964), Lamonds (1972), MacDonald et al. (1977), Imes (1990a, b), Smith and Imes (1991), Christenson et al. (1994), Imes and Emmett (1994), Bell et al. (1996), and Jorgensen et al. (1996). Regional stratigraphic studies were conducted by Heller (1954), Knight (1954), Sheldon (1954), Caplan (1957, 1960), Snyder (1976), Thompson (1991, 1995), and McFarland (1998).

The area of investigation is bounded by Missouri on the north, Oklahoma on the west, the boundary between the Interior Highlands and the Gulf Coastal Plain provinces on the east, and the Arkansas Baseline (a line separating north Townships from south Townships) on the south, and includes northern and central Arkansas (figure 1). The investigation focuses on the Ozark Plateaus province of Arkansas because hydrologic data exist for only Baxter, Benton, Boone, Carroll, Fulton, Izard, Lawrence, Madison, Marion, Newton, Randolph, Searcy, Sharp, Stone, and Washington Counties. Limited geologic

information is included for the southern portion of the study area.

The study area is located in the Interior Highlands of Arkansas, which is topographically higher and more rugged than the Gulf Coastal Plain. The Interior Highlands occupies about 48 percent of Arkansas and is divided into three major physiographic regions: the Ozark Plateaus, Arkansas Valley, and Ouachita Mountains (figure 2).

The Ozark Plateaus province is subdivided into the Salem Plateau, the Springfield Plateau, and the Boston Mountains. The Salem Plateau extends southward from Missouri and is topographically the most subdued. It is underlain dominantly by Ordovician dolostones and interbedded sandstones. The Springfield Plateau lies between the Salem Plateau and the Boston Mountains. It is intermediate in elevation and ruggedness, and is generally underlain by a sequence of limestone and chert of Mississippian age. The Boston Mountains are the southernmost and most rugged portion of the region. The mountains are capped by sandstones and shales of Pennsylvanian age (Howard et al., 1997).

The Ozark Plateaus province of Arkansas has a temperate climate and ranges from 42 to 48 inches of annual precipitation (figure 3a; Imes and Emmett, 1994). Estimated mean annual evapotranspiration rates are 30 to 35 inches (Adamski et al., 1995). The recharge area for the Roubidoux Formation and Gunter Sandstone Member is in southern Missouri, where the mean annual recharge ranges from 5 to 15 inches per year (figure 3b; Jorgensen et al., 1996).

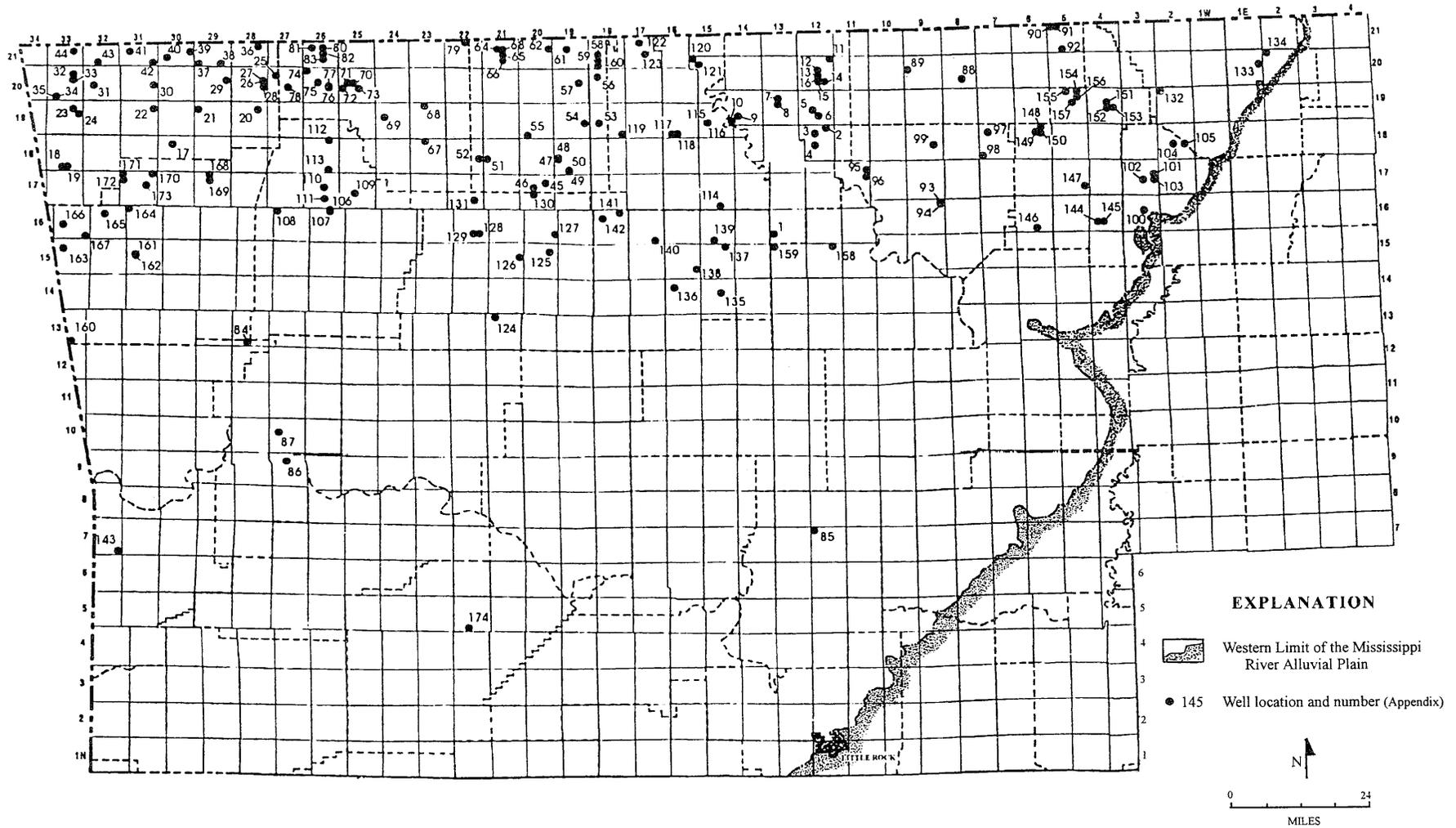


Figure 1. Location of wells included in this study.

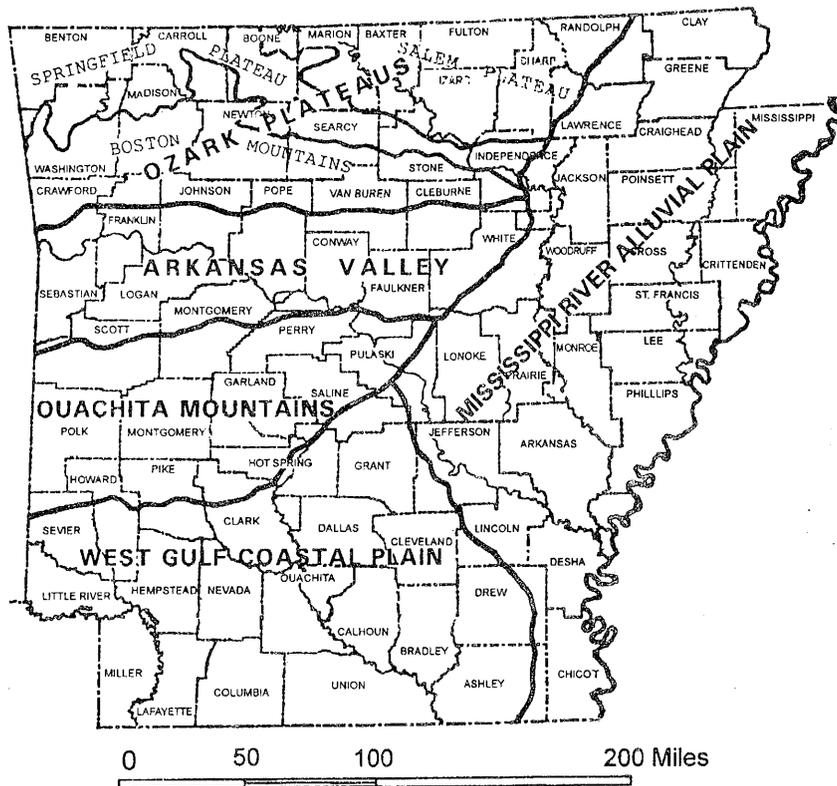


Figure 2. Physiographic provinces of Arkansas
(Modified from Howard, Colton, and Prior, 1997)

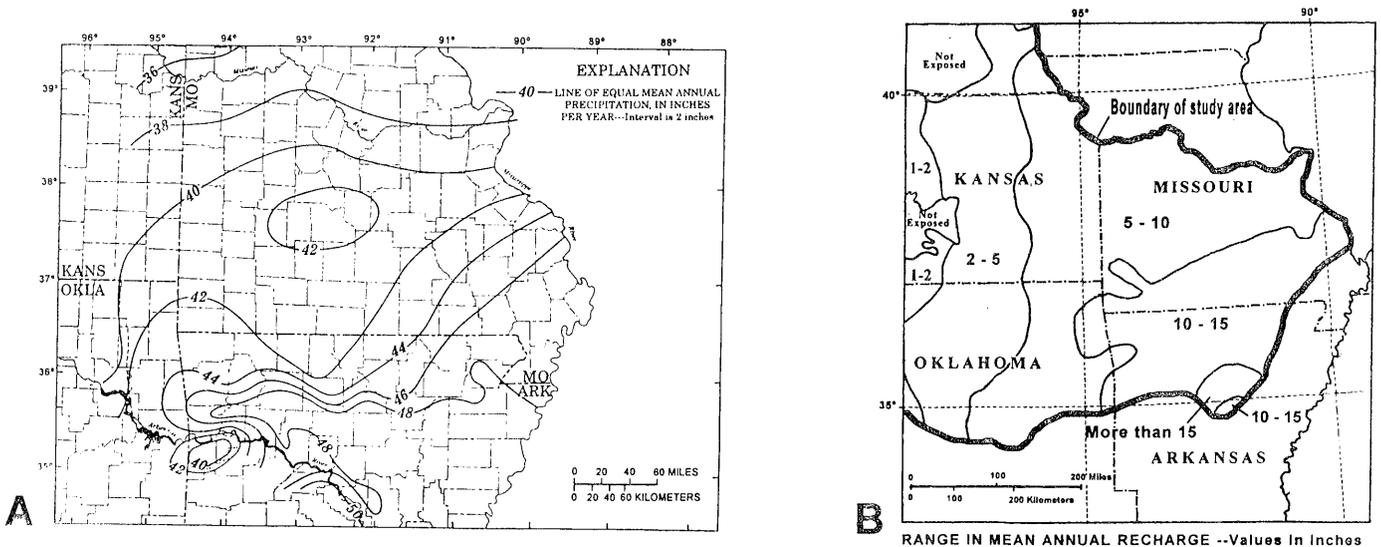


Figure 3A. Mean annual precipitation of the Ozark Plateaus region. (Data from Imes and Emmett, 1994, figure 4)

Figure 3B. Mean annual recharge of the Ozark Plateaus region. (Data from Jorgensen et al., 1996, figure 5)

Period	Formation	Hydrologic Unit		
CARBONIFEROUS	PENNSYLVANIAN	Atoka	Boston Mountains/Western Interior Plains Confining Unit	
		Bloyd		
	Hale	Prairie Grove		
		Cane Hill		
	MISSISSIPPIAN	Pitkin		
		Fayetteville		
		Batesville		
		Moorefield		
		Boone		Springfield Plateau Aquifer
		St. Joe		
DEVONIAN	Chattanooga	Ozark Confining Unit		
	Clifty	Ozark Aquifer		
	Penters			
SILURIAN	Lafferty			
	St. Clair			
ORDOVICIAN	Cason		Brassfield	
			Late	Fernvale
	Kimmswick			
	Plattin			
	Middle		Joachim	
			St. Peter	
		Everton		
		Early	Powell	
			Cotter	
			Jefferson City	
	Roubidoux			
	Gasconade			
	CAMBRIAN	Gunter	St. Francois Confining Unit	
Eminence				
Potosi				
Derby-Doerun-Davis				
Bonnerterre		St. Francois Aquifer		
Lamotte				
PRECAMBRIAN	Igneous rocks	Basement Confining Unit		

FIGURE 4. Correlation chart of Arkansas formations and Ozark Plateaus hydrologic units. No relative thickness is implied by this chart.

Geology

The Ozark region is a broad structural and sedimentary platform surrounding the Precambrian igneous rocks that compose the St. Francois Mountains in southeastern Missouri. In Arkansas, the Precambrian igneous basement is buried by the overlying sedimentary rocks of Late Cambrian through Early Pennsylvanian age, which form the Ozark Plateaus province. The Roubidoux Formation and Gunter Sandstone Member of the Gasconade Formation are Early Ordovician in age (figure 4, modified from J. D. McFarland, 1998, and USGS Water Resources Division data).

The Roubidoux unconformably overlies the Gasconade and is composed of cherty dolostone, dolomitic sandstone, and scattered sandstone intervals. The dolostone of the Roubidoux is finely- to coarsely-crystalline and light gray to brown in color. The formation's thickness ranges from 130 to 260 feet in the area where hydrologic data is available and thickens to the southeast in Conway, Faulkner, and White Counties to over 300 feet. The formation is thinnest in Benton and Washington Counties. Local variation in thickness is apparently due to irregularities on the surface on which it was deposited (MacDonald et al., 1977). Outcrops are present in a broad area in southeastern Missouri. The Roubidoux is conformably overlain by the Jefferson City Dolomite (MacDonald et al., 1977).

The Gasconade unconformably overlies the Late Cambrian Eminence Formation (figure 4). It is predominantly a light brownish-gray, cherty dolostone. Except in Howell, Oregon, and Ozark Counties in southeastern Missouri (T. L. Thompson, personal communication), the Gasconade contains a persistent

sandstone unit in its lowermost part that is designated the Gunter Sandstone Member. The upper portion of the formation is dominantly medium-crystalline dolostone and contains relatively small amounts of chert. The part of the formation below the "upper" portion and above the Gunter Sandstone Member is generally coarsely crystalline and may contain up to 50 percent chert by volume. The Gasconade ranges in thickness from 300 to 600 feet in northern Arkansas. Outcrops are present in a broad area in southeastern Missouri.

The Gunter Sandstone Member of the Gasconade is composed of quartzose sandstone and sandy dolostone, except in those counties in Missouri mentioned above where it consists of a few thin sand beds or "floating" sand grains in the dolostone succession. In Arkansas, the Gunter is a recognizable and persistent subsurface unit, which lies unconformably above the Eminence Formation. It ranges in thickness from 20 to over 100 feet and averages about 30 feet. The area of greatest thickness includes southern Boone and northern Newton Counties and has been attributed to irregularities in the underlying Eminence Formation (Caplan, 1960). Gunter outcrops are confined to a narrow belt in southeastern Missouri.

The restricted number of wells drilled to and through the Roubidoux and Gunter limits subsurface investigations in northern Arkansas. Caplan (1960) and Lamonds (1972) both used 28 wells as data points to construct contour maps on selected formations. MacDonald et al. (1977) used 48 wells in a study of the Roubidoux and Gasconade Formations. The data set of this report contains 174 wells, although no map uses all wells in the data set. Previous investigators eval-

uated lithology, electric and gamma logs, and insoluble residue patterns to delineate the tops of formations, and thus formation thickness. The presence, tops, and thickness of the Roubidoux and the Gunter for the additional wells studied for this report (Appendix) were evaluated by examination of well cuttings. In the predominantly dolostone sequence, the presence or absence of sand grains, sandstone, and sandy cherts was used to define the upper and lower contacts of both aquifers. Well distribution is influenced by local area needs for water supply. Northwest Arkansas has a greater density of water wells drilled to these aquifers than the remainder of the Ozark Plateaus province (figure 1).

The elevations of the top of the Roubidoux and Gunter relative to mean sea level are shown by contour lines in Figures 5 and 6, respectively. Given a specific well site elevation, depth to the upper contacts of the units may be calculated. The contour maps show a general southerly dip for both aquifers. The dip increases near the -1000-foot elevation on the Roubidoux Formation (figure 5) and the -1500-foot elevation on the Gunter Sandstone (figure 6). Sharp bends in the contours may be related to local structures, such as displacement across faults, as shown in Boone, Newton, and Searcy Counties (figures 7a and 7b). Well and chemical data are too sparse to evaluate possible effects of regional post-lithification structure (faulting) plotted on Figures 7a and 7b. However, both surface mapping and inferred subsurface faulting suggest a general lack of faulting influence on the aquifers in northern Carroll, Boone, and Marion Counties and in Fulton, Randolph, Sharp, and IZard Counties. Additional subsurface mapping during this

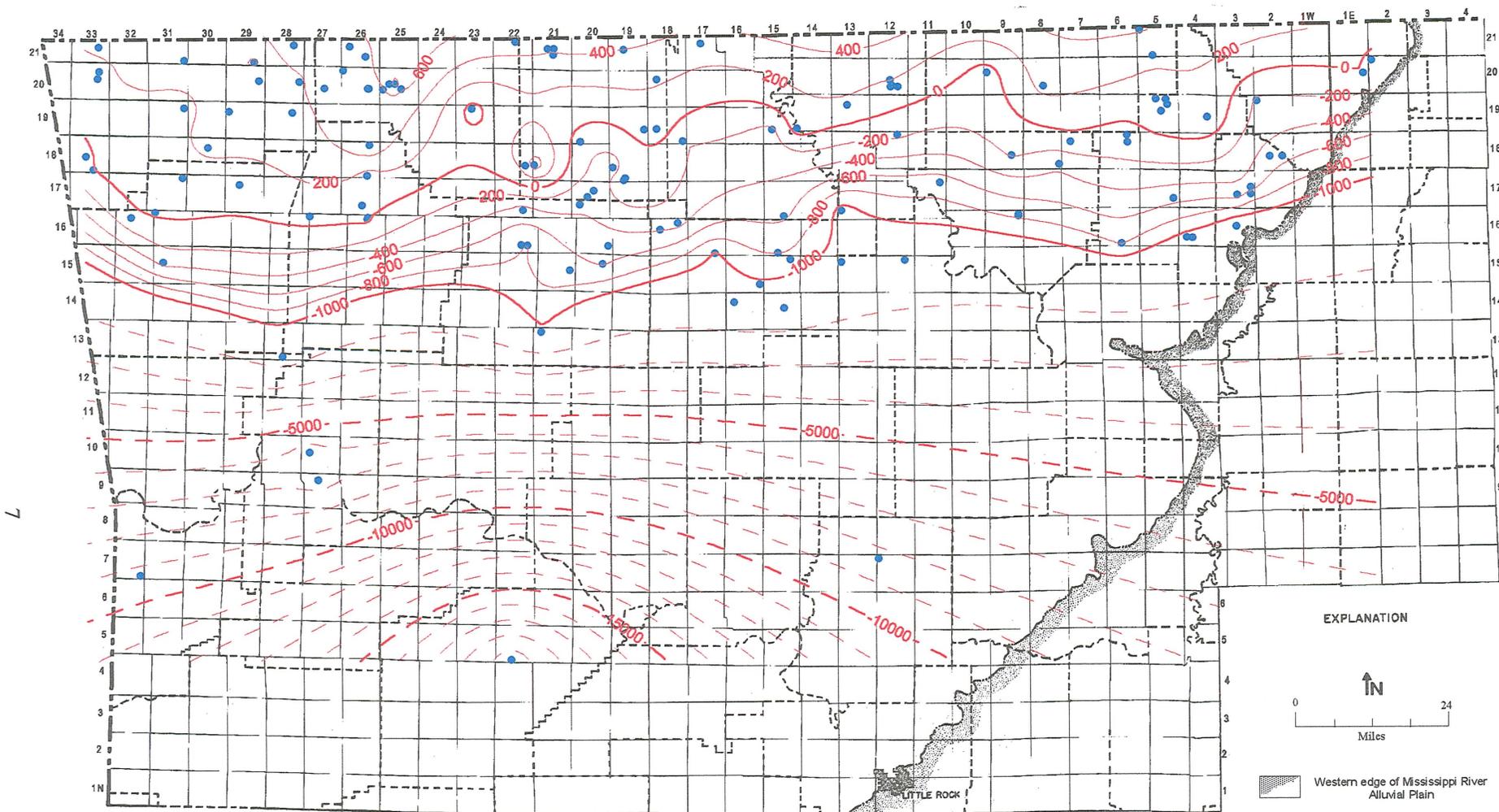


Figure 5. Structure contour map of the top of the Roubidoux Formation

The blue dots mark the locations of the wells used to construct this map. The contour interval is 200 feet above -1000 feet and 1000 feet below -1000. This map was created in Surfer Surface Mapping System (Version 6.02) software from Golden Software, Inc. The data set was grided for contouring by the Kriging method (with a linear variogram) utilizing an extrapolated grid density of 288 lines east-west and 144 lines north-south.

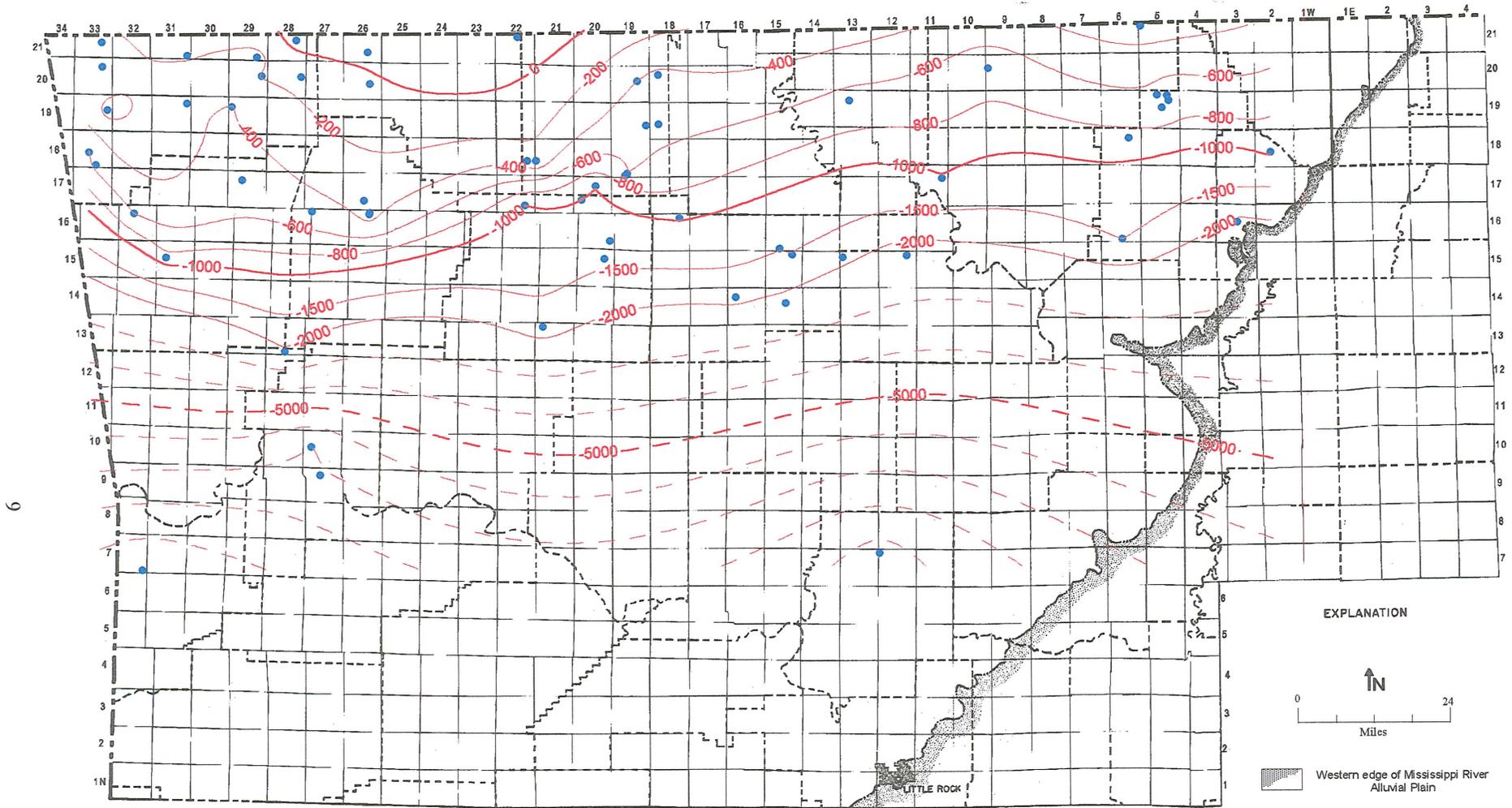


Figure 6. Structure contour map of the top of the Gunter Sandstone Member of the Gasconade Formation

The blue dots mark the locations of the wells used to construct this map. The contour interval is 200 feet above -1000 feet, 500 feet between -1000 and -2000 feet, and 1000 feet below -2000 feet. The map was created in Surfer Surface Mapping System (Version 6.02) software from Golden Software, Inc. The data set was grided for contouring by the Kriging method (with a linear variogram) utilizing an extrapolated grid density of 288 lines east-west and 144 lines north-south.

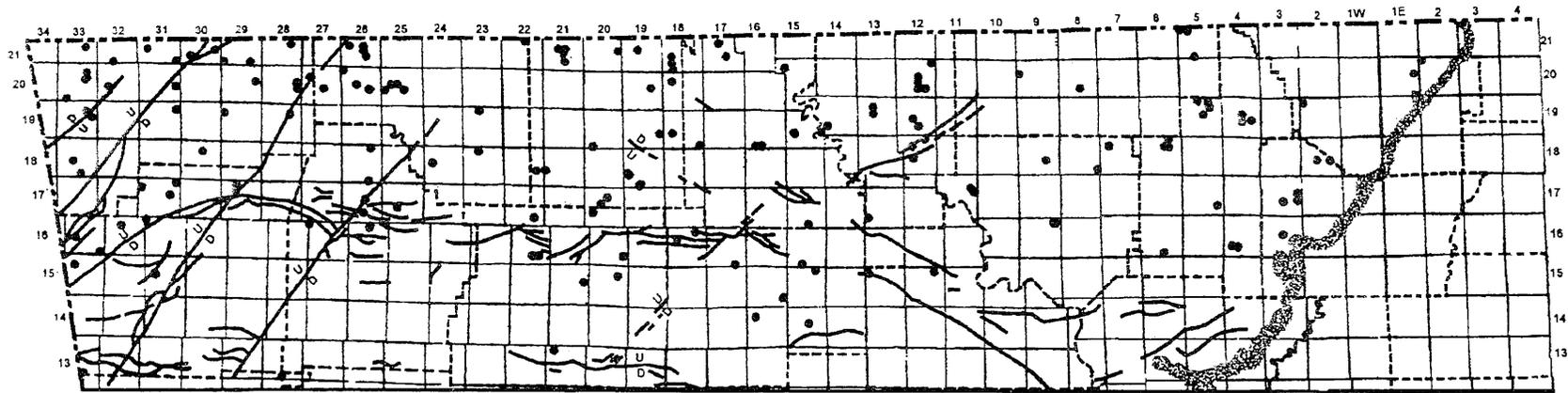


Figure 7A. Surface faults of northern Arkansas. (Data from Haley, et al., 1993.)

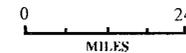
EXPLANATION



Western Limit of the Mississippi River Alluvial Plain



Relative direction of fault displacement



11

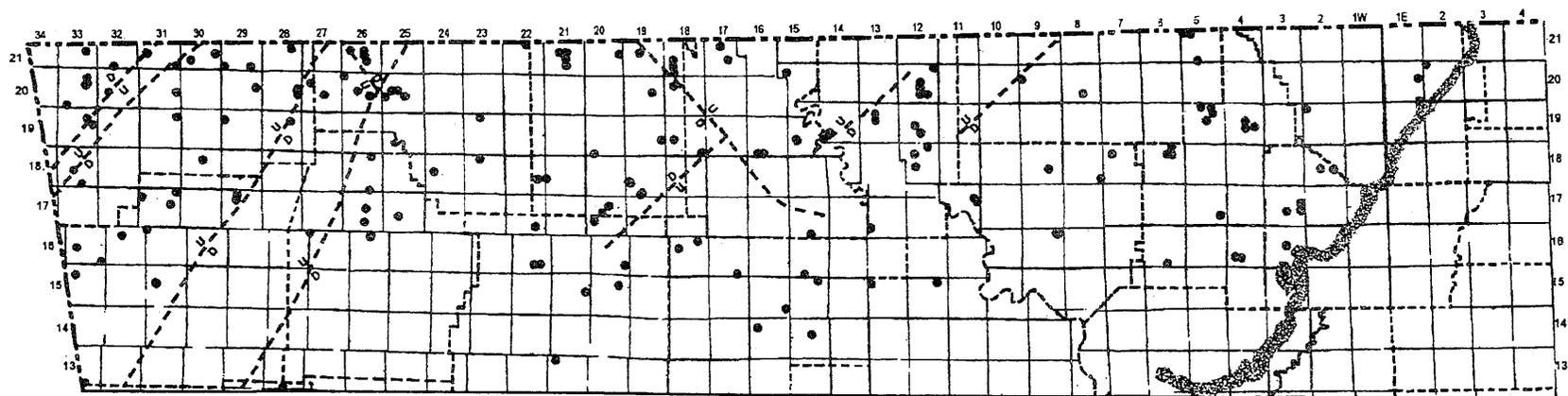


Figure 7B. Inferred faults that may influence the Roubidoux - Gunter Strata. (Data from Plate 2, MacDonald, et al., 1977.)

project was deemed not feasible due to an insufficient number of wells. Broad structural highs in western Carroll and Benton Counties (figures 5 and 6) are caused by irregularities in the Precambrian basement (MacDonald et al., 1977). Smaller structural highs and lows in the Ozark Plateaus province may indicate local solution, compaction, and subsidence.

Only wells where the top and bottom of the aquifer could be located were used to produce isopach (thickness) maps of the units (figures 8 and 9). These maps may be used to evaluate the depth and thickness of the major water-producing units. Unconformities at the base of the Roubidoux and Gunter may affect their thickness (MacDonald et al., 1977). The apparent thickness of the Roubidoux as indicated by this investigation (figure 8) seems to vary only over broad areas.

Hydrogeology

This report used water levels and well yields as the principal parameters for hydrogeologic evaluation, with a focus on the areal distribution of these parameters. According to the U. S. Geological Survey (Imes and Emmett, 1994), the Roubidoux Formation and Gunter Sandstone member comprise only a small part of the Ozark Aquifer (figure 4). In Missouri, many other Lower Ordovician formations produce ground water. Recharge is largely from the outcrop areas in southern Missouri (figure 3b) and water flow follows the regional dip of the sedimentary units toward the south. Some Missouri wells drilled into the Roubidoux and Gunter utilize an unknown percentage of the recharge before it can reach Arkansas.

In Arkansas, water in the Roubidoux and Gunter is confined between impervious strata and is under hydrostatic pressure from a column of water extending upward to the recharge area. Under these conditions, water in a well will rise to a level above the top of the highest water-producing unit if the well is not being over pumped. Ideally the amount of rise is measured during a test period when the well is not being pumped. This is the static water level. Elevations of static water level are shown in Figure 10. The map does not represent a true potentiometric surface for either of the aquifers, but it may be used as a guide to estimate the minimum depth for pump installation.

Well yields in the Roubidoux and Gunter are affected by the following factors: depth of penetration of the well bore into the water-producing unit, total thickness, relative proportion of sandstone to dolostone, degree of cementation of sandstone, and secondary porosity/ permeability from recrystallization of the dolostone. Well yields from the Roubidoux in Arkansas ranges from 20 to 600 gallons per minute (GPM) (Figure 11). The wide range in yield may be caused by several factors: 1) some wells penetrate the entire section of the Roubidoux, while others do not; 2) local variation in the porosity and permeability due to differing cementation and the presence or absence of clay in the unit; 3) variable thickness of the Roubidoux; or 4) proximity to faults. When insufficient yield is obtained from the Roubidoux, wells are drilled into the underlying Gunter. Well yields in the Gunter Sandstone are usually combined for both aquifers (figure 12). Wells producing from both units range in yields from 26 to 502 gpm.

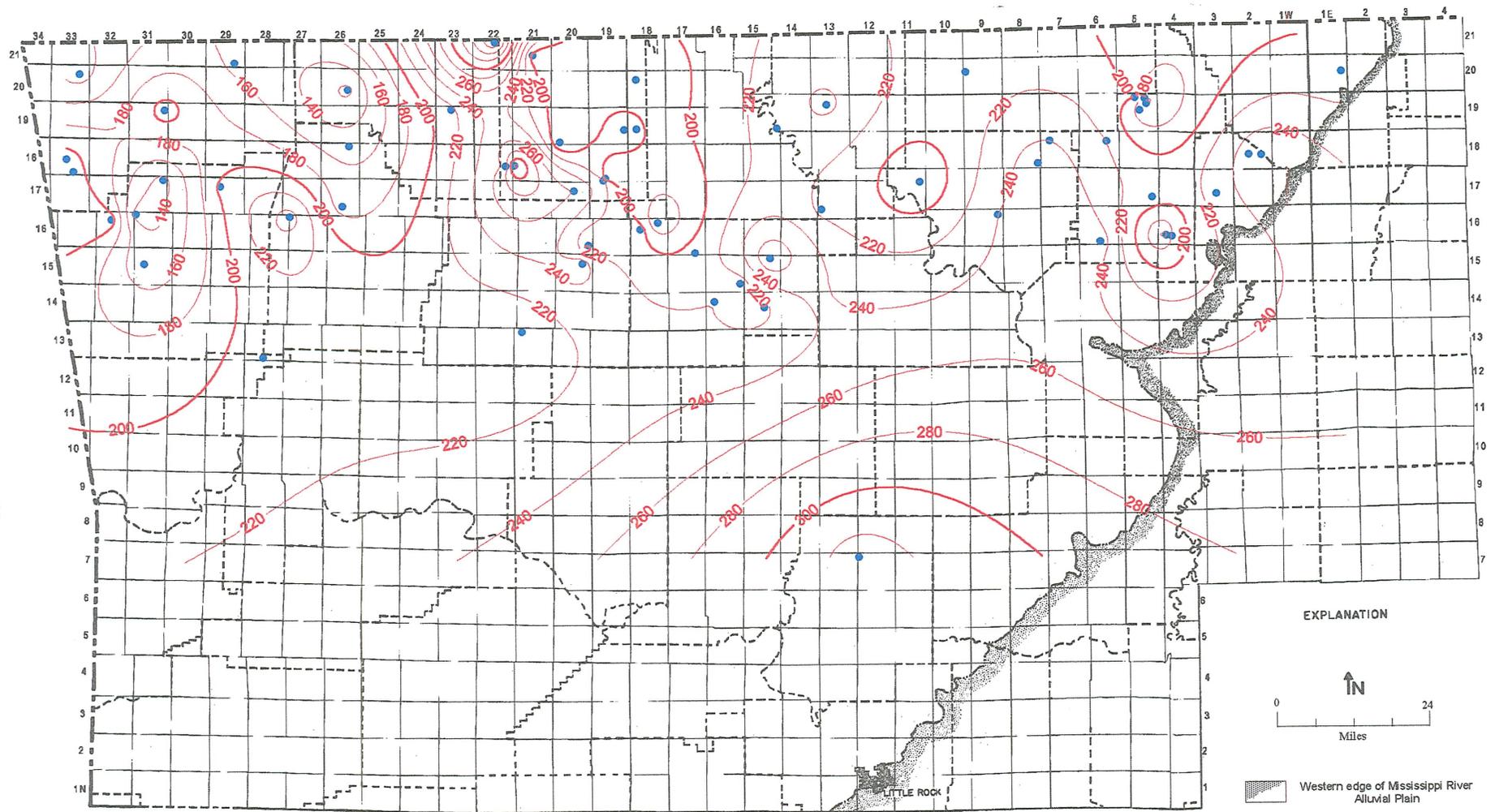


Figure 8. Isopach (thickness) map of the Roubidoux Formation

The blue dots mark the locations of the wells used to construct this map. The contour interval is 20 feet. The map was created in Surfer Surface Mapping System (Version 6.02) software from Golden Software, Inc. The data set was grided for contouring by the Kriging method (with a linear variogram) utilizing an extrapolated grid density of 288 lines east-west and 144 lines north-south.

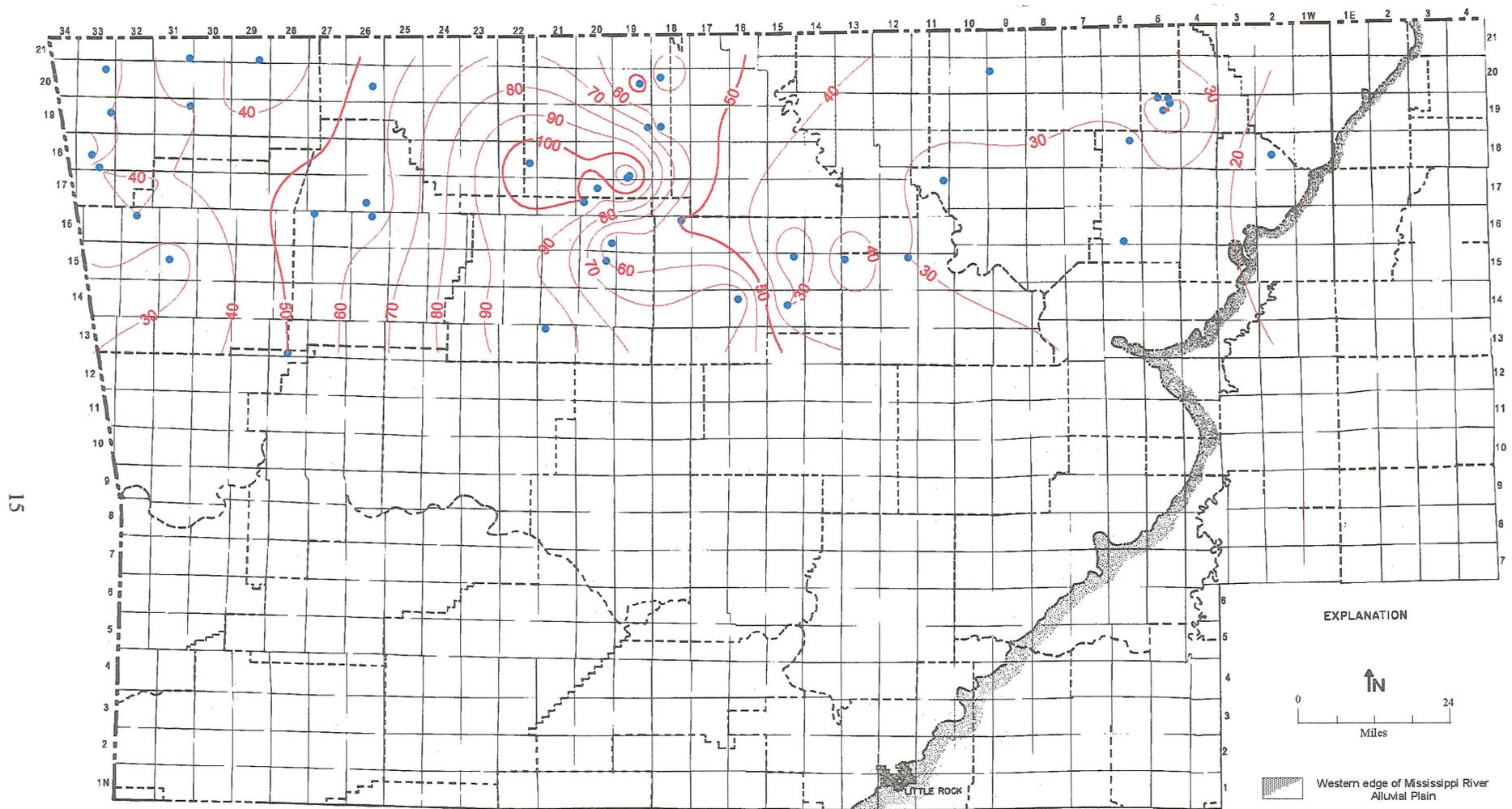


Figure 9. Isopach (thickness) map of the Gunter Sandstone Member of the Gasconade Formation

The blue dots mark the locations of the wells used to construct this map. The contour interval is 10 feet. The map was created in Surfer Surface Mapping System (Version 6.02) software from Golden Software, Inc. The data set was gridded for contouring by the Kriging method (with a linear variogram) utilizing an extrapolated grid density of 288 lines east-west and 144 lines north-south.

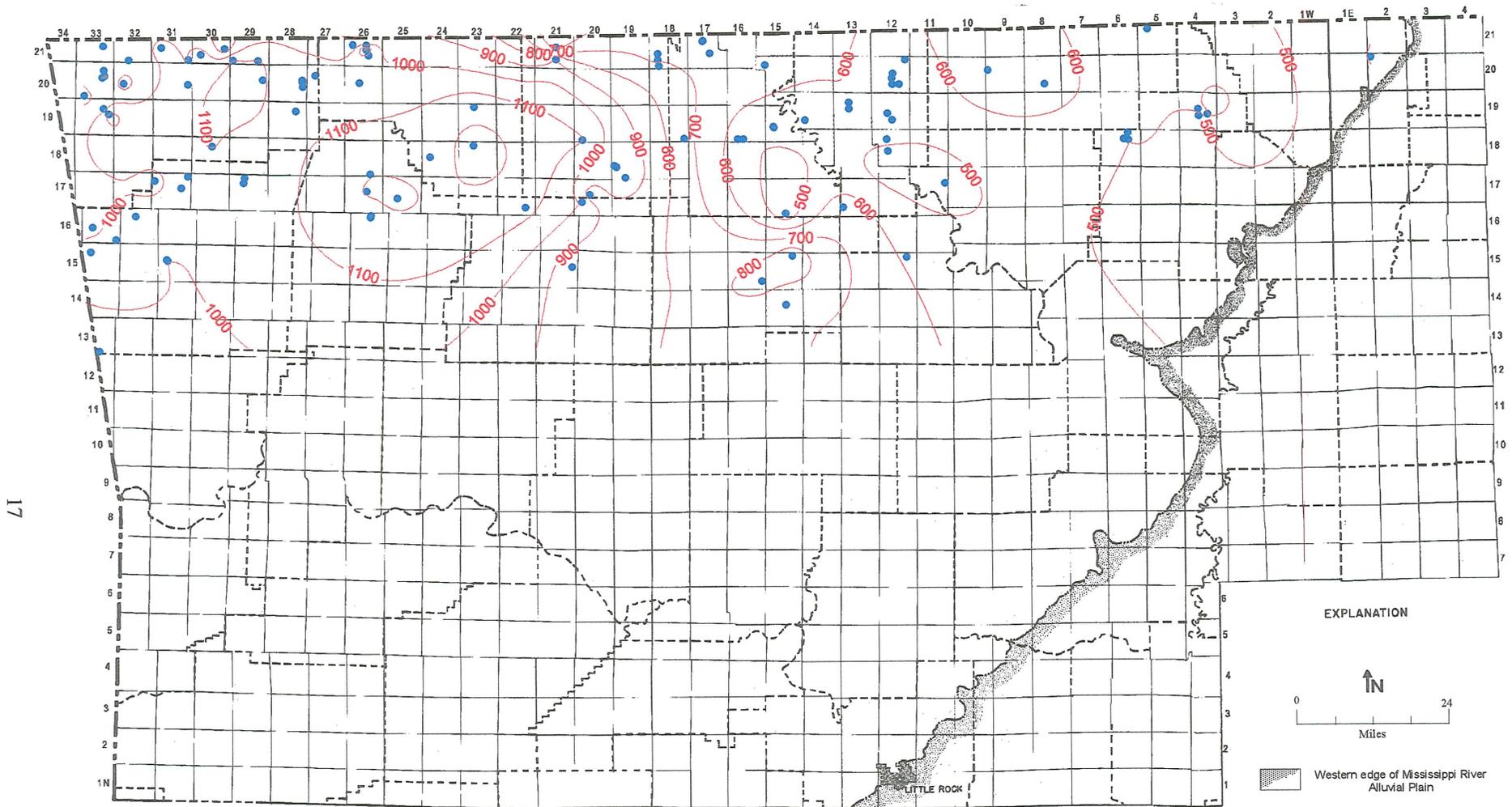


Figure 10. Elevation (feet above sea level) of the static water level in selected water wells.

The blue dots mark the locations of the wells used to construct this map. The contour interval is 100 feet. The map was created in Surfer Surface Mapping System (Version 6.02) software from Golden Software, Inc. The data set was gridded for contouring by the Kriging method (with a linear variogram) utilizing an extrapolated grid density of 288 lines east-west and 144 lines north-south.

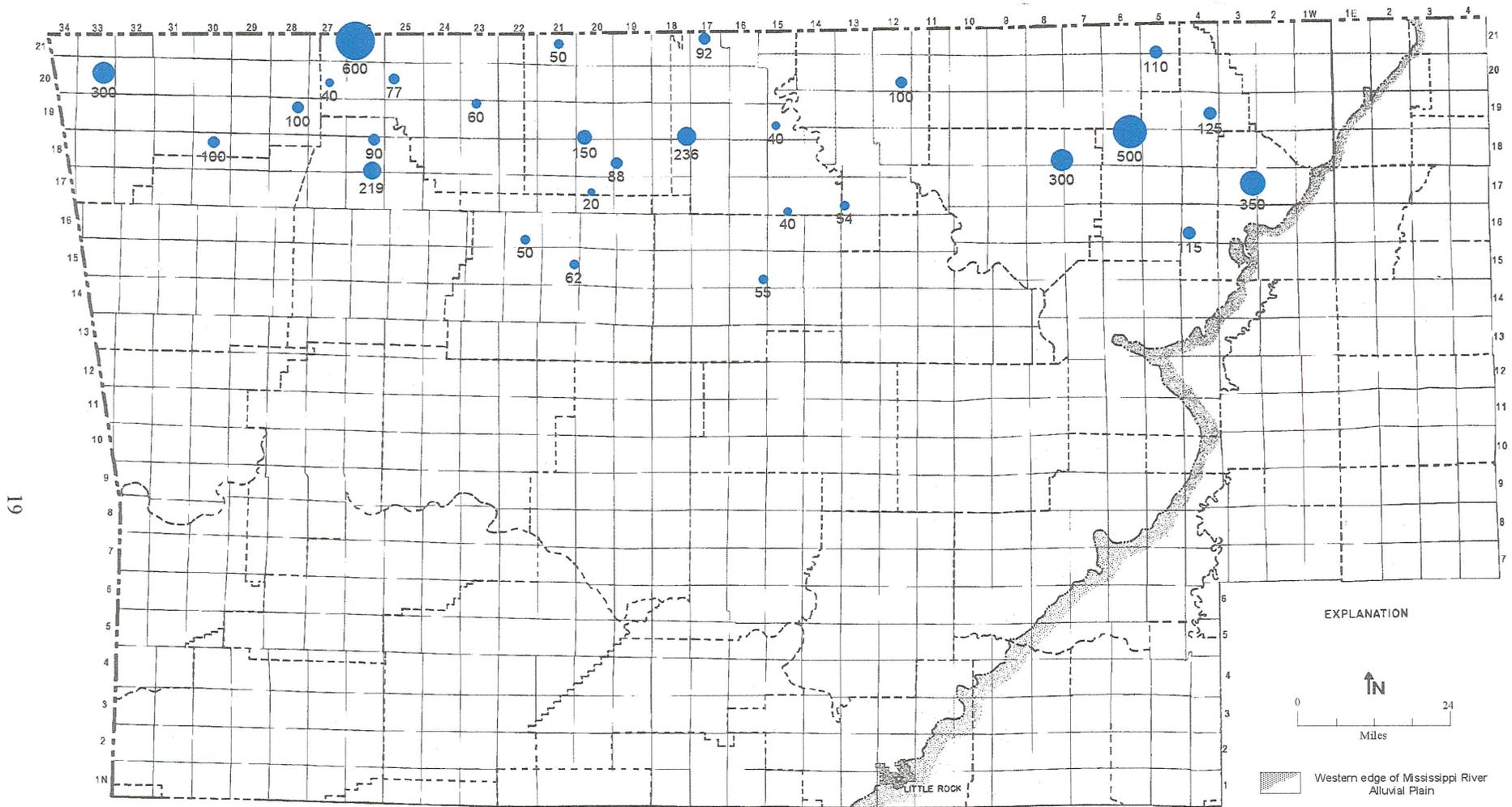
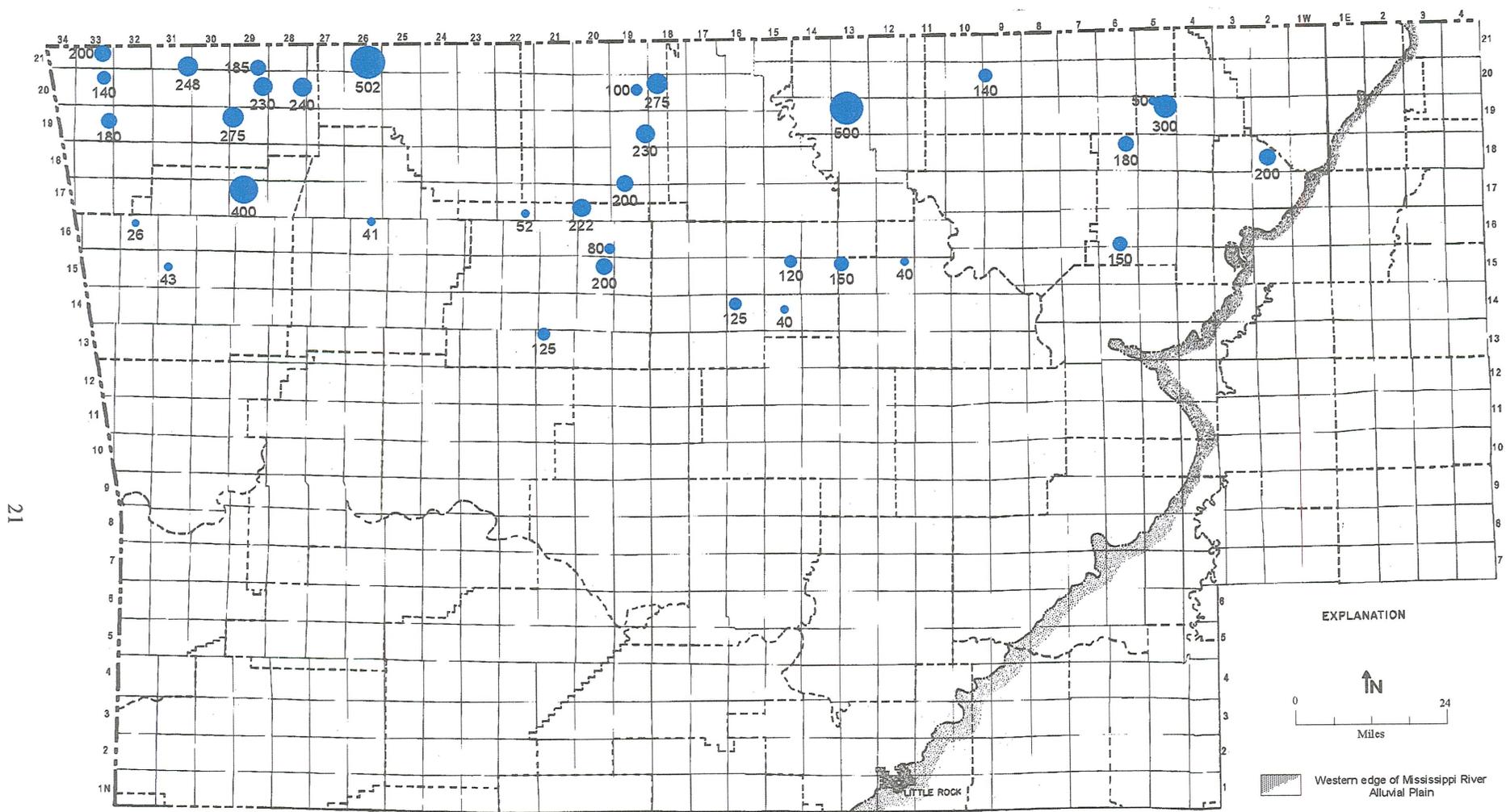


Figure 11. Yields (gpm) of selected water wells producing from the Roubidoux Formation



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Figure 12. Yields (gpm) of selected water wells producing from both the Roubidoux Formation and the Gunter Sandstone Member

Water Quality

A set of water chemistry data is presented in the Appendix. Water quality is defined by the National Drinking Water Standards, which is based on the maximum concentrations of constituents considered to be safe in drinking water. Table 1 shows selected National Secondary Drinking Water Standards (NSDWS) as set by the U. S. Environmental Protection Agency and adopted by the Arkansas Department of Health. The Arkansas Department of Health requires treatment of ground water that does not meet these standards.

pH	6.5-8.5
Total Dissolved Solids	500 mg/L
Chloride	250 mg/L
Iron	0.3 mg/L
Sulfate	250 mg/L

Table 1. Selected National Secondary Drinking Water Standards (source: Arkansas Department of Health)

Christenson et al. (1994) report the presence of brine in the Roubidoux Formation in western northeast Oklahoma and an intermediate zone of mixing of the fresh water and brine. The southern end of the mixed zone trends east toward Arkansas. Jorgensen et al. (1996) show a zone of high chloride concentration in Oklahoma entering Arkansas from the west in the vicinity of Fort Smith and trending east across the Arkansas Valley region.

Only a few water wells have been drilled to the Roubidoux south of Township 14 North. One of these, the Nail-Swain #1 well (124, Appendix), was relatively high in chloride and contained a high value of fluoride (16.3 mg/L). The NSDWS for fluoride is 4 mg/L. South of

the Boston Mountains in the northern Arkansas Valley, most wells reaching the Roubidoux were drilled as test holes for oil or gas production. No specific water analyses are available for these wells, but they were reported to contain brine. No evaluation of possible zones of mixed waters in the Roubidoux and Gunter on the southern margin of the Arkansas Ozark Plateaus (similar to that documented in Oklahoma) is possible due to insufficient data. If deeper wells are drilled in the extreme southern Ozark Plateaus province, they probably will encounter higher chloride concentrations and increasing water salinity. The combination of increasing salinity southward across the Ozark Plateaus and increased drilling costs may limit the southward extent of drilling to the Roubidoux and Gunter in search of potable water. At present, this limit has not been defined.

Water Analyses

Chemical analyses of waters from the Roubidoux Formation and Gunter Sandstone Member of the Gasconade Formation are tabulated in the Appendix. No analyses exist for water solely from the Gunter, but instead represent the mixing of waters from both formations.

Sources of Water Data

Water analyses in this report are primarily from the U. S. Geological Survey. Some of the wells were sampled several times over a multi-year span, while others were sampled only once. Where multiple samples were available, the earliest analyses were used. This approach creates a uniform data set and minimizes the effects of water-quality changes due to pumping over time. When earlier analyses were incomplete,

the earliest analysis was used. Additional analyses were obtained from MacDonald et al. (1977) and the Arkansas Department of Health.

Chemical Data

Despite numerous studies, the chemical data set on wells completed in the Roubidoux and Gunter is relatively small. Due to cost and physical limitations, most analytical work has been conducted on municipal wells and much less on commercial and privately owned wells. If the well yield is sufficient, only a small portion of the Roubidoux may be penetrated. If well yield is insufficient, the drilling will often be extended to the Gunter. Only 60 wells in the study area have extensive water analyses. Of these wells, only 28 represent water from the Roubidoux. Most of the data set (32 wells) consists of analyses of the mixed formation waters produced from both aquifers. In either case, it is possible that the samples are mixed due to infiltration of waters from units higher in the section.

An interpretation of chemical data in the Appendix should be done with caution due to the sparseness of the data compared to the areal extent of the study area. In many instances, only one analytical value is available for a given well. Several wells have multiple analyses which show changes in water chemistry with pumping over time. Initial analyses may represent highly localized ground water conditions in a given formation or contaminants from the drilling and pump-setting stages of well development. The general trend of water chemistry of wells analyzed periodically over time indicates an improvement in the water quality as defined by the National Drinking Water Standards.

Temperature, pH, and total alkalinity were measured in the field. All other chemical data are laboratory analyses and/or calculated values (G. Stanton, personal communication). Data collected in the field may be subject to variations in test procedures by the many workers involved during the period of investigation (1950-1995). Ideally, for instance, the pH and temperature are measured on a sample from a well after sufficient pumping removes 3 times the well's storage capacity (G. Stanton, personal communication). However, this procedure may not have been followed by every analyst at every well.

Laboratory analyses may also change over time due to evolving methodologies and techniques as well as changes in the well. The use of more sensitive analytical equipment has resulted in the lowering of detection limits of many water-soluble chemical constituents.

Chemical data with 3 decimals (0.00X) may not be accurate because of uncertainty with the third decimal (Hem, 1992), but was the only analysis available when included in the Appendix. Many of the chemical analyses listed in the Appendix were not originally expressed past the second decimal. A value of <0.003 or 0.001 should only be interpreted as indicating very low, but detectable values. In most cases, these values represent the detection limits of the analytical method (G. Stanton, personal communication). Accuracy of analytical results for major constituents of water is ± 2 to ± 10 percent (Hem, 1992). This indicates a difference of 2 to 10 percent between the reported result and the true sample concentration at the time of analysis. Concentrations greater than 100 mg/L are generally accurate to with-

in ± 5 percent. Precipitation on the collection container wall after sample collection may alter water chemistry. Consequently, the chemical data in the Appendix should be interpreted as a best approximation of the true value.

Temperature

Temperature of ground water normally increases with depth due to the geothermal gradient. Shallow ground-water temperature (shallow wells and springs) usually reflects the average yearly temperature of the region. In the study area, shallow ground-water temperature is 18.2°C (65°F), based on the average of 27 springs in the study area. In general, the temperature increases about 1°C for each 100 meters (1.8°F for each 328 feet) in depth in north Arkansas. The following temperatures at given depths were reported on a temperature log run by Southeast Region Logger Service for the USGS on the Nail-Swain #1 well in Newton County: 22°C at 354 meters (71.6°F at 1160 feet), 23°C at 451 meters (73.4°F at 1480 feet), 24°C at 518 meters (75.2°F at 1700 feet), 25°C at 628 meters (77°F at 2060 feet), 26°C at 726 meters (78.8°F at 2380 feet), 28°C at 902 meters (82.4°F at 2959 feet), 29°C at 1018 meters (84.2°F at 3340 feet), 30.8°C at 1183 meters (87.4°F at 3880 feet).

Results

Total Dissolved Solids

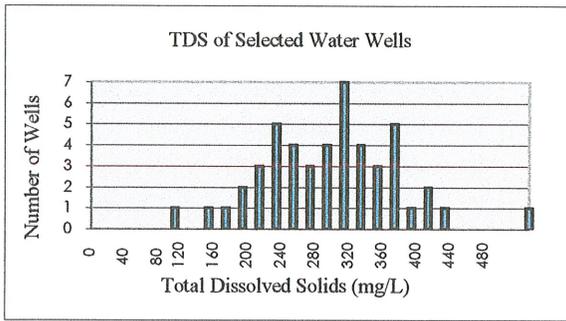
The total concentration of dissolved material in water is ordinarily determined from the weight of the dry residue remaining after evaporation of the volatile portion of a specified volume of the water sample. This value is called

the total dissolved solids (TDS). If the concentration of the major ions have been determined, the TDS may be calculated by summing the constituents. One flaw with the evaporative method is that water is retained by some residue types. However, the residue left by evaporation can usually be used to verify a calculated value. In the Appendix, the TDS concentrations were determined by summation of the constituents.

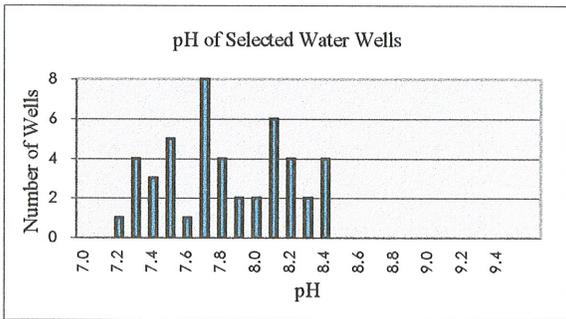
TDS concentrations (figure 13) range from 93 to 685 mg/L with a mean of 278 mg/L. Only one value was above the NSDWS of 500 mg/L TDS. The highest concentration was from the Nail-Swain #1 well (well 124) in Newton County, the southernmost water well reaching the Roubidoux Formation and Gunter Sandstone. A map plot of selected wells for TDS (figure 14) shows no significant pattern to the concentration distribution.

pH

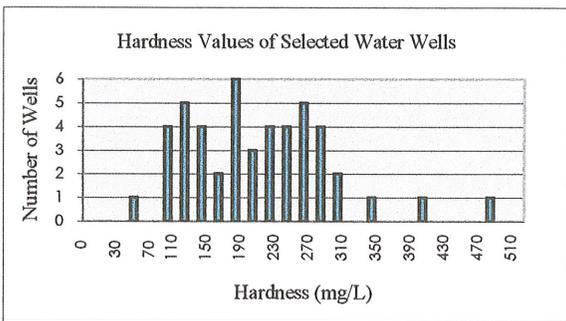
The pH of water is the measure of its acidity or alkalinity. A pH value indicates concentrations of H^+ and OH^- in solution. The pH of natural waters may vary widely from acid (<7.0) to alkaline (>7.0). The pH is influenced by mineral dissolution in the formation(s) in which water resides, elevated concentrations of CO_2 in the soils through which the water originally passed, and the source of the water (atmospheric, formational, or brines). Acidic rainwater reacts with carbonate rocks upon entry into the subsurface in the recharge area in southern Missouri, resulting in slight alkalinity. Additionally, the carbonate cements of the Roubidoux and Gunter may further alter the ground water chemistry. Once the water reaches a near neutral pH, then adsorption



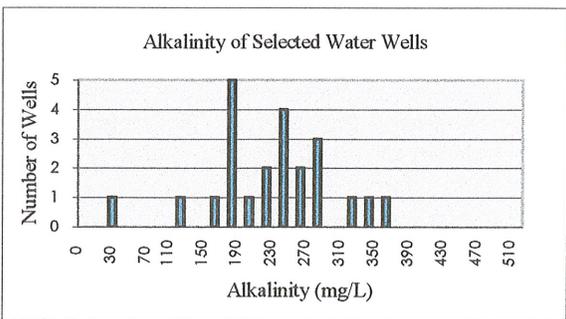
TDS
 No. samples 48
 Max. 685
 Min. 93
 Mean 278.56
 Standard deviation 93.04



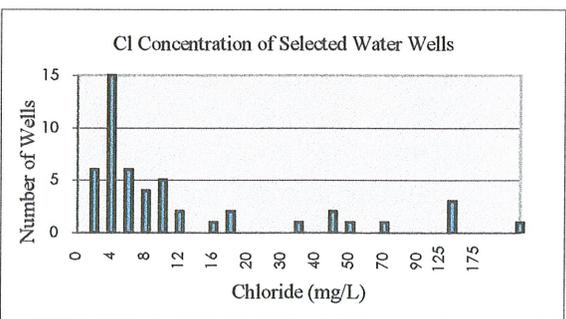
pH
 No. samples 46
 Max. 8.4
 Min. 7.2
 Mean 7.82
 Standard deviation 0.35



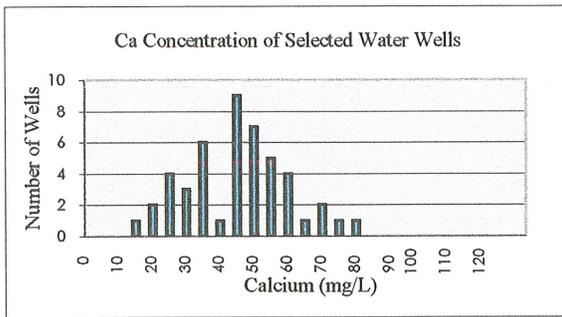
Hardness
 No. samples 47
 Max. 470
 Min. 51
 Mean 200.83
 Standard deviation 84.12



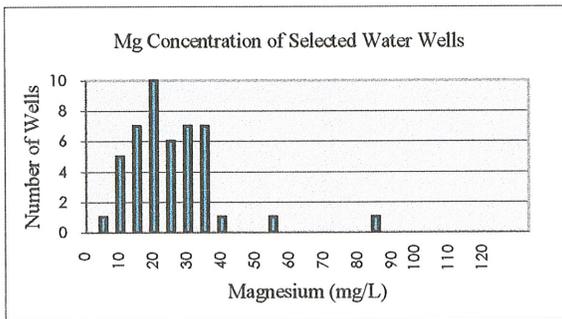
Alkalinity
 No. samples 23
 Max. 354
 Min. 38
 Mean 219.00
 Standard deviation 71.06



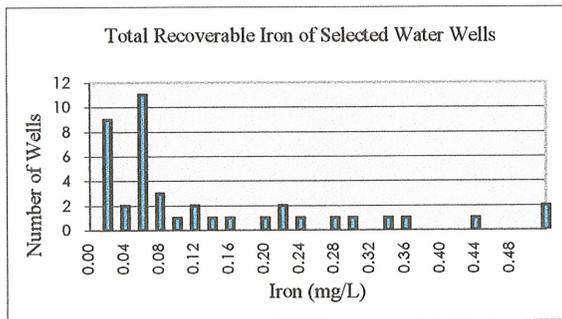
Cl
 No. samples 50
 Max. 240
 Min. 1.4
 Mean 21.04
 Standard deviation 42.38



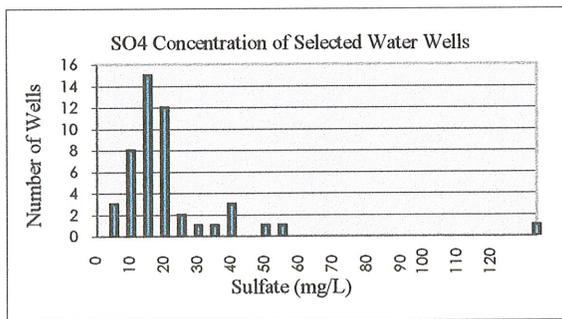
Ca	
No. samples	47
Max.	79
Min.	11
Mean	43.55
Standard deviation	15.30



Mg	
No. samples	46
Max.	83
Min.	1
Mean	23.06
Standard deviation	13.47



Recoverable Iron	
No. samples	41
Max.	3.6
Min.	0.001
Mean	0.214
Standard deviation	0.574



SO4	
No. samples	48
Max.	131
Min.	0.4
Mean	19.12
Standard deviation	19.95

Figure 13. Histograms of data from selected wells.

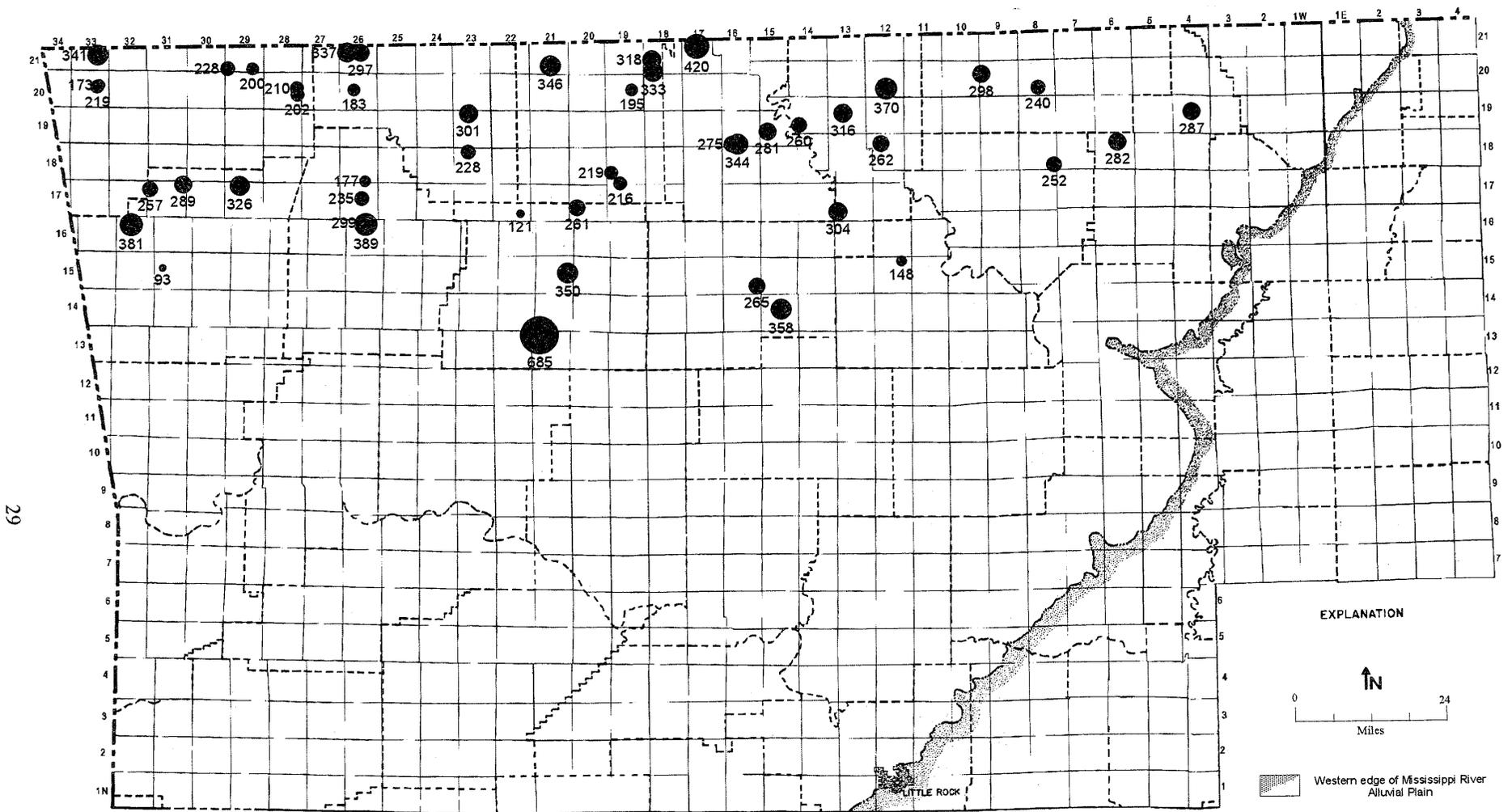


Figure 14. Total dissolved solids (TDS) of selected water wells

of certain dissolved ions on grain surfaces may occur because of negative charges on grain surfaces (Hem, 1992). Some effects on ground water chemistry may only be local.

The slightly alkaline pH values of the water data (7.2 to 8.4) are typical for carbonate-sourced ground water and are within the NSDWS for pH (table 1). The histogram for pH (figure 13) shows a mean of 7.82.

Hardness

The term “hardness”, although commonly used, is difficult to define. In the past, hardness was evaluated by effects observed in the use of soap or with encrustations left by some types of water after heating. If reaction with soap was only considered, then hardness represents the soap-consuming capacity of water. Since most of the effect observed with soap comes from the presence of calcium and magnesium, hardness is now generally defined in terms of these elements alone, with possible interferences (American Society for Testing and Materials, 1964, p. 391). The modern method for analysis of hardness, developed in the 1940’s, utilizes individual instrumental element analysis and is much more reliable than the soap method. Since hardness is caused by the presence of several elements, a convention is used to express concentrations in quantitative terms. Usually, hardness values are reported as equivalent concentrations of calcium carbonate. The term “total hardness” is calculated by taking the sums of calcium and magnesium concentrations (expressed as mg/L) and multiplying by 50. As a result, a practical definition of hardness becomes “the effect of alkaline-earth cations” (Hem, 1992). The hardness

range has been classified by Durfor and Becker (1964):

- 0-60 mg/L – Soft
- 61-120 mg/L – Moderately hard
- 121-180 mg/L – Hard
- Greater than 180 mg/L – Very hard

In domestic water, hardness is not objectionable until it reaches a concentration of about 100 mg/L. Where waters have been in contact with limestone or gypsum, hardnesses of 200 to 300 mg/L or greater may occur.

Hardness concentrations in this study range from 51 to 470 mg/L with a mean of 201 mg/L (figure 13). About one half of the water samples from the Roubidoux Formation and Gunter Sandstone are moderately hard to very hard. The highest concentration reported is from the City of Lead Hill well (well 60) in Boone County.

Alkalinity

The alkalinity of a solution is its capacity to react with and neutralize acid and is influenced by carbon dioxide (CO₂) content. Alkalinity is a calculated parameter accomplished through a field procedure that uses titration. Acid is added to the water sample until the solution’s pH reaches a predetermined pH value. Natural waters attain alkalinity during water-rock interaction. Alkalinity usually indicates only bicarbonate and carbonate concentration, given in mg/L. The sources of CO₂ species in both surface and ground water are: 1) the CO₂ gas fraction in the atmosphere; 2) the CO₂ gas fraction in atmospheric gases in soil; and 3) gases in the zone below the soil horizon and above the water table (vadose zone). Carbon dioxide in ground water may be increas-

ed during respiration of vegetation and oxidation of organic matter (Hem, 1992).

Alkalinity concentrations (figure 13) range from 38 to 354 mg/L with a mean of 219 mg/L. The highest concentration reported is from the Corps of Engineers well (well 120) in Marion County.

Chloride

Chlorine is the most common of the halogen elements and may exist in some minerals (such as halite) and as a gas. It is readily soluble in water where it exists as Cl^- (chloride). Chloride is present in all natural ground waters derived from rainfall, but with low concentrations. Both sodium and chloride are present in high concentrations in brines and sea water, resulting in salinity.

The data set presented in this study meets the NSDWS maximum limit for chloride of 250 mg/L (Table 1). Chloride concentration (figure 13) ranges from 1.4 to 240 mg/L with a mean of 21 mg/L. The data set is positively skewed toward values less than 12 mg/L. The highest concentration reported is from the Hugh W. Webb well (well 37) in Benton County, although it is well within the NSDWS (table 1). A plot of the chloride concentration of selected wells (figure 15) shows higher chloride concentrations in the western portion of the study area.

Calcium

Calcium is the most abundant of the alkaline-earth metals. It is a major constituent of many rock-forming minerals and a major component of the total dissolved solids in most natural waters. The most common calcium-bearing minerals in sedimentary rocks are calcite and aragonite (calcium carbonate), dolo-

mite (calcium magnesium carbonate), and gypsum (calcium sulfate hydrate). In sandstones and other detrital rocks, calcium carbonate is common as a cement or as a partial filling of interstices between grains. Calcium is also present in the form of adsorbed ions on negatively charged mineral surfaces in soils, sediments, and rocks. Carbonate rocks (limestone/dolostone) in the recharge area and carbonate cements in the formations all influence calcium concentration. The dominant source providing calcium to the ground water in the Roubidoux and Gunter is unknown, but it is likely the carbonate-dominated section of rocks composing the recharge area in southern Missouri.

The concentrations of calcium range from 11 to 79 mg/L with a mean of 44 mg/L (figure 13).

Magnesium

Calcium and magnesium contribute similarly to the hardness of water. However, magnesium ions are considerably smaller than sodium or calcium, which leads to precipitation of magnesium compounds. The source of magnesium in ground water is commonly the mineral dolomite, which composes sedimentary beds of dolostone. Magnesium also occurs in high concentrations in most limestones. Dissolution of both dolostones and limestones releases magnesium into solution, but the process is not readily reversible. Therefore, the magnesium concentration increases along the flow path of groundwater undergoing such processes, until a high Mg:Ca ratio is achieved. The general characteristics of ground water from dolomitic terrain indicate that saturation relative to CaCO_3 also occurs in these waters. The dominant source providing magnesium to

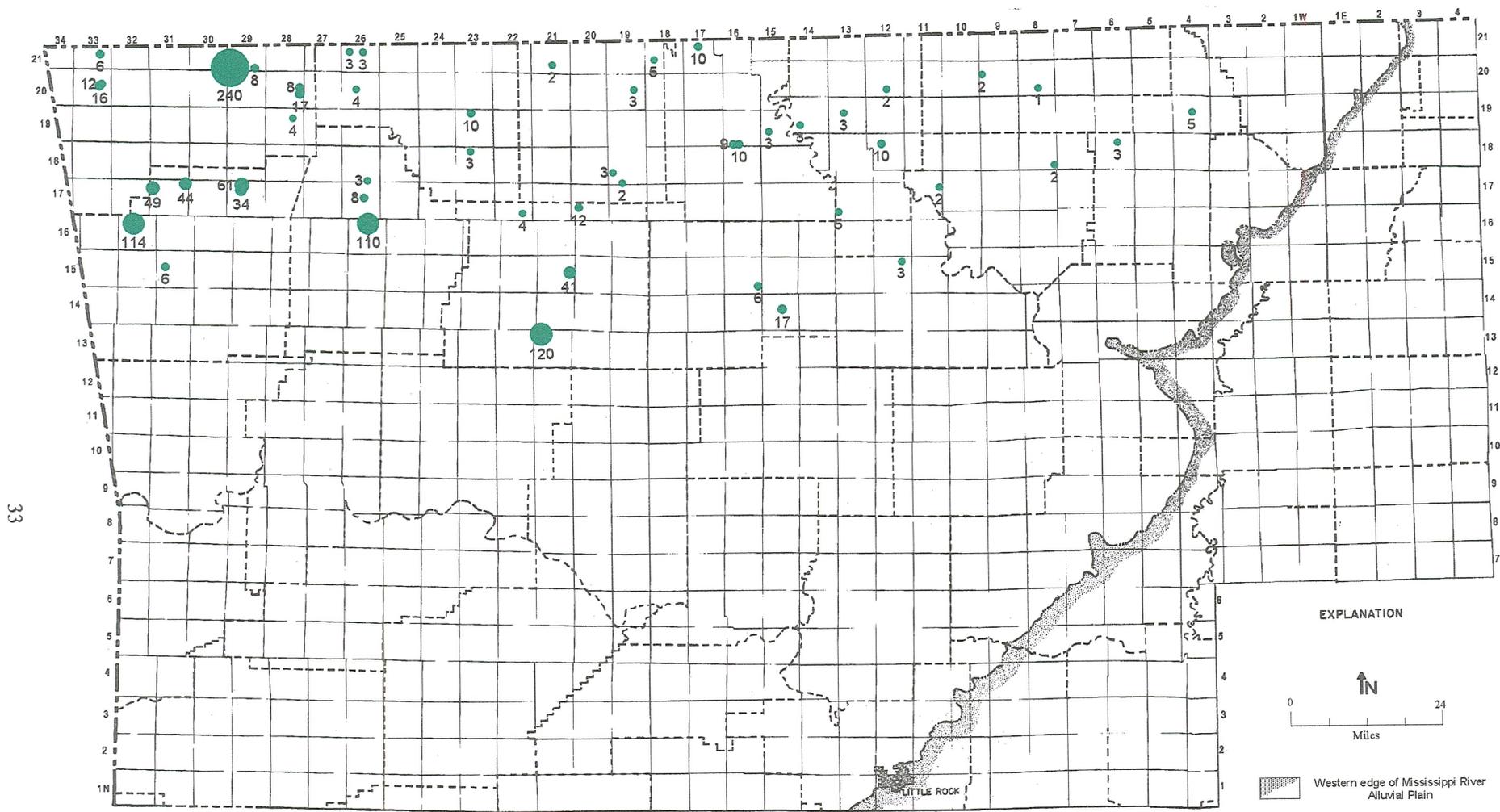


Figure 15. Chloride concentration (mg/L) of selected water wells

ground water in the Roubidoux and Gunter is unknown, but is likely the carbonate-dominated section of rocks in the recharge area of southern Missouri.

The concentrations of magnesium in this study range from 1 to 83 mg/L with a mean of 23 mg/L (figure 13).

Iron

Total iron concentration in this study was originally calculated in micrograms/liter by the USGS analysts. Their values were converted to milligrams/liter (mg/L) because chemical constituent concentrations listed in the National Drinking Water Standards are given in mg/L, as are the concentrations reported in the Appendix.

Dissolved iron in ground water comes from the dissolution of iron-bearing minerals during water-rock interaction. Soluble iron-bearing minerals in sedimentary rocks are pyrite or marcasite (iron sulfide), limonite (iron hydroxide), or hematite or goethite (iron oxides). Iron may exist in water in both the ferric or ferrous state. If the system contains sufficient oxygen, ferrous iron is oxidized to ferric iron.

Concentrations reported represent total recoverable iron. The NSDWS for iron is 0.3 mg/L. In general, water quality for both the Roubidoux and Gunter meets these standards; only 4 wells in the data set were above 0.3 mg/L. Iron concentration ranged from 0.001 to 3.6 mg/L with a mean of 0.214

mg/L (figure 13). The histogram shows a positive skewness of values above 0.110. When iron concentrations are above 0.3 mg/L, municipal water treatment facilities can remove the iron by treatment.

Sulfate

The presence of sulfur in the form of hydrogen sulfide (H_2S) gas in ground water creates a “rotten egg” odor. The presence of sulfate (SO_4) does not create an odor, but results in “soft water”. Soft water creates difficulty in the removal of soap film. Sulfate in ground water comes from the dissolution of sulfur-bearing minerals, such as pyrite-marcasite (iron sulfide), gypsum (calcium sulfate), or glauberite (sodium calcium sulfate). Additionally, sulfur is released during aerobic bacterial action on organic matter in unconsolidated sediments and sedimentary rocks above the water table in an oxygen-rich environment. After being released, sulfur may readily oxidize to sulfate or, in the presence of hydrogen, be converted to H_2S .

All the analyses in the study’s data set meet the NSDWS maximum limit for sulfate of 250 mg/L (Table 1). Sulfate concentrations range from 0.4 to 131 mg/L with a mean of 19 mg/L (figure 13). The histogram displays a positive skewness for higher concentrations, but all values are within the drinking water standard.

Acknowledgments

The authors are indebted to many people for their cooperation and assistance in providing information concerning geology and hydrogeology of the Roubidoux Formation and the Gunter Sandstone Member of the Gasconade Formation. The cooperation of the U. S. Geological Survey -Water Resources Division and the Arkansas Department of Health was crucial in providing hydrologic and water chemistry information, without which this study could not have been completed. Outside reviews of the manuscript by G. W. Colton, Dr. D. L. Zachry, and D. A. Freiwald improved the content and focus of the report markedly. Special assistance from Susan Young, Cartography Supervisor of the Arkansas Geological Commission, is greatly appreciated.

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Appendix

Water Wells Used in the Compilation of This Report

The Elev. is the surface elevation at the well head. TD is the total depth of the well. The Rbx Top and Gtr Top list the elevations of the top of the Roubidoux Formation and the Gunter Sandstone Member of the Gasconade Formation respectively. Rbx Thick and Gtr Thick list thickness values for the Roubidoux and Gunter respectively. WL is the elevation of the static water level in the well. Yield is the reported production of the well in gallons per minute (gpm). TDS (total dissolved solids), Alkal (alkalinity), Cl, Ca, Mg, Fe, and SO₄ are expressed as milligrams per liter (mg/L).

Well	County	Well Name	Location	Elev.	TD	Rbx Top	Rbx Thick	Gtr Top	Gtr Thick	W L	Yield	TDS	pH	Hardness	Alkal	Cl	Ca	Mg	Fe	SO ₄	Temp (C)	
1	Baxter	Big Flat #1	T16N,R13W, Sec30	1300	2603	-1080	215			660	54	304	7.7	180	270	4.5	30.0	19.0	0.050	52.0		
2	Baxter	Norfork Dam	T18N,R12W, Sec02	640	910	-160																
3	Baxter	Salesville	T18N,R12W,Sec09	725	835					595	170	262		210	198	10.0	51.0	21.0	0.350	7.0		
4	Baxter	Norfork	T18N,R12W,Sec21	500	1507					409	122											
5	Baxter	Corps Engs.	T19N,R12W,Sec21	662	652					565	45											
6	Baxter	Corps Engs.	T19N,R12W,Sec27	645	652					569	110											
7	Baxter	Mtn Home #3	T19N,R13W,Sec09	809	1562	154	245	-671		601	500											
8	Baxter	Mtn Home #1	T19N,R13W,Sec16	740	1540					564	185	316	7.7	290		3.0	60.0	35.0	3.600	14.0		
9	Baxter	City of Cotter #2	T19N,R14W,Sec29	720	1625					543	400	260	7.5	260		2.8	50.0	33.0		14.0	19.0	
10	Baxter	City of Colter	T19N,R14W,Sec31	800	1090	80	230															
11	Baxter	Gamaliel Land	T20N,R12W,Sec01	705	606					570	110											
12	Baxter	Corps Engs.	T20N,R12W,Sec15	652	511					557	100											
13	Baxter	Bidwell Point	T20N,R12W,Sec22	650	547	210				566												
14	Baxter	Henderson #1	T20N,R12W,Sec26	605	510	135				536	100											
15	Baxter	Panther Bay #1	T20N,R12W,Sec27	665	550	120				546	100	370	8.0	400	38	2.0	75.0	52.0	0.300	13.0		
16	Baxter	Panther Bay #2	T20N,R12W,Sec27	600	490	120					145											
17	Benton	Billy Acres Sbd	T18N,R30W,Sec09	1295	1160	235				1104	100											
18	Benton	EW Curry #1	T18N,R33W,Sec33	1138	2236	-52	210	-607	15													
19	Benton	Ozark Prod.1	T18N,R33W,Sec34	1190	2222	-5	205	-550	40													
20	Benton	Rocky Branch #1	T19N,R28W,Sec11	1260	1030	280				1068	100		8.1	110	262	4.0	43.0	1.0	0.040	18.0		

Well	County	Well Name	Location	Elev.	TD	Rbx Top	Rbx Thick	Gtr Top	Gtr Thick	W L	Yield	TDS	pH	Hard-ness	Aikal	Cl	Ca	Mg	Fe	SO4	Temp (C)	
21	Benton	Cty of Rogers	T19N,R29W,Sec07	1220	1659	350		-405		1080	275											
22	Benton	Misp-Grdy Jns1	T19N,R31W,Sec12	1273	2338	143	210	-372	40													
23	Benton	Decatur	T19N,R33W,Sec11	1200	1720					942	350											
24	Benton	Cty of Decatur	T19N,R33W,Sec13	1275	1450			-140	30	1139	180											
25	Benton	Mor Loch	T20N,R27W,Sec08	1295	1875					1056	20											
26	Benton	Lst Brdge Vil #1	T20N,R28W,Sec13	1380	1626	360		-145		1039	240	210	7.9	170		7.5	37.0	18.0	0.050	16.0	19.5	
27	Benton	Corps Engs.	T20N,R28W,Sec13	1438	1037					1031	70											
28	Benton	Lst Brdg Village	T20N,R28W,Sec24	1380	1150					1028	300	202	8.4	190	160	17.0	43.0	21.0	0.050	13.0		
29	Benton	Avoca Water D #1	T20N,R29W,Sec13	1432	1968	310		-190		1107	230											
30	Benton	Jebco Inc	T20N,R31W,Sec24	1120	1100					1066	120											
31	Benton	JE Farriester	T20N,R32W,Sec20	1220	1170					854	40											
32	Benton	Cty of Gravette2	T20N,R33W,Sec11	1210	1600	139	130	-361	25	960	140											
33	Benton	Cty of Gravette3	T20N,R33W,Sec14	1230	1614	195				1000	300	219	8.2			16.0				8.0		
34	Benton	Cty of Gravette1	T20N,R33W,Sec14	1200	1603					1030	125	173	7.4	110		12.0	25.0	11.0		14.0		
35	Benton	Monte Shaha	T20N,R33W,Sec32	1165	1133					873	8											
36	Benton	Gateway #1	T21N,R28W,Sec14	1615	1610	555		65														
37	Benton	Hugh W Webb	T21N,R29W,Sec31	1300	1274					1115	50	228		190	175	240.0	42.0	20.0	0.020	11.0		
38	Benton	Pea Ridge NPS	T21N,R29W,Sec35	1406	1769	171	155	-259	30	1108	185	200	7.8	180	170	8.0	42.0	19.0	0.270	14.0	20.0	
39	Benton	Bella Vista	T21N,R30W,Sec24	1220	2000					825	319											
40	Benton	Bella Vista	T21N,R30W,Sec29	1260	1985					1048	299											
41	Benton	Bella Vista	T21N,R31W,Sec20	1242	1115					1025	50											
42	Benton	Lambeth Bella V	T21N,R31W,Sec36	1260	2000	290		-180	50	995	248											
43	Benton	Don Tyson	T21N,R32W,Sec33	1280	1156					938	60											
44	Benton	Sulpher Sprgs #1	T21N,R33W,Sec23	1036	1442	240		-260		917	200	341	7.8	260	222	5.5	64.0	26.0	0.230	39.0		
45	Boone	Krooked Creek	T17N,R20W,Sec11	1245	2370	-275	260	-995	90													
46	Boone	Big Rck Cdy Mtn	T17N,R20W,Sec16	1250	1800	-380				850	20											
47	Boone	Cty of Bellefont	T18N,R19W,Sec19	1150	1649					1010	96	219	8.1	160	170	2.5	33.0	20.0	0.050	18.0		
48	Boone	AHTD	T18N,R19W,Sec19	1050	2000	-320				926	88											
49	Boone	Valley Sprg Mt 1	T18N,R19W,Sec33	1300	2055	-20	230	-710	120	967	200	216	7.7	190	170	1.5	41.0	22.0	0.050	20.0	19.5	

Well	County	Well Name	Location	Elev.	TD	Rbx Top	Rbx Thick	Gtr Top	Gtr Thick	WL	Weld	TDS	pH	Hardness	Alkal	Cl	Ca	Mg	Fe	SO4	Temp (C)	
50	Boone	Valley Sprg Mt 2	T18N,R19W,Sec33	1330	2050	10	190	-600	120		225											
51	Boone	Barrett #1	T18N,R21W,Sec19	1470	2246	520	330	-250														
52	Boone	RHEA Bryan1	T18N,R22W,Sec24	1894	2343	274	265	-326	110													
53	Boone	Prowett #1	T19N,R18W,Sec20	920	2646	123	203	-545	54													
54	Boone	Cty of Bergman 2	T19N,R19W,Sec24	1400	2090	70	220	-570	70		230											
55	Boone	Lindsey Const #1	T19N,R20W,Sec32	1187	1505	-113	180			1107	150											
56	Boone	City of Lead Hill	T20N,R18W,Sec08	895	1420	225	180	-435	67		275											
57	Boone	Paul Epps 7A	T20N,R19W,Sec14	760	2412			-403	44		100	195	7.7	180	179	2.8	41.0	20.0	1.100	7.7	16.0	
58	Boone	H W Dietz	T21N,R18W,Sec20	830	604					673		318	8.1	250		5.0	49.0	30.0	0.070	20.0		
59	Boone	H W Dietz	T21N,R18W,Sec29	750	703					664	650											
60	Boone	Cty of Lead Hill	T21N,R18W,Sec32	840	602					741		333	7.3	470			50.0	83.0	0.430	11.0		
61	Boone	Tucker Hollow R	T21N,R19W,Sec16	700		400																
62	Boone	Corps Engs	T21N,R20W,Sec13	710	348						35											
63	Boone	Unknown	T21N,R21W,Sec15	1340	847	530				525	50											
64	Boone	Rivera St Line P	T21N,R21W,Sec16	1290	868	480					35											
65	Boone	Omaha #2	T21N,R21W,Sec22	1300	1340	310	190				115											
66	Boone	Cty of Omaha	T21N,R21W,Sec27	1357	1315					907	52	346	8.1	210		2.0	45.0	25.0	0.070	40.0		
67	Carroll	Howard Collins	T18N,R23W,Sec04	1310	1032					1238		228	7.5	220	211	2.7	48.0	25.0	0.010	13.0	16.0	
68	Carroll	Cty of Grn Fores	T19N,R23W,Sec04	1350	1587	-66	215			1164	60	301	8.0	250	228	10.0	54.0	28.0	0.150	32.0		
69	Carroll	W J Schell	T19N,R24W,Sec17	1410	1314					1146	150											
70	Carroll	Riverside Res.	T20N,R25W,Sec16	1287	506	897																
71	Carroll	Usrey Mobil Pk	T20N,R25W,Sec17	1160	560	690					77											
72	Carroll	Longview Realty	T20N,R25W,Sec19	1420	890	580					100											
73	Carroll	Osage Mobil Pk	T20N,R25W,Sec22	1200	780	610																
74	Carroll	Rivertake OutDor	T20N,R26W,Sec06	1420	1003	460																
75	Carroll	Eureka Sprgs	T20N,R26W,Sec16	1200	1332					1022	500	183	8.2	170		4.0	34.0	16.0		18.0		
76	Carroll	Cty of Eureka Sp	T20N,R26W,Sec23	1334	1713	519	115	-31	55													
77	Carroll	Cty of Eureka Sp	T20N,R26W,Sec23	1330	902	460																
78	Carroll	Greencliff Subd	T20N,R27W,Sec22	1350	870	550					40											

Well	County	Well Name	Location	Elev.	TD	Rbx Top	Rbx Thick	Gtr Top	Gtr Thick	WL	Yield	TDS	pH	Hardness	Aikal	Cl	Ca	Mg	Fe	SO4	Temp (C)	
79	Carroll	Branson RV Pk	T21N,R22W,Sec10	1164	1200	504	380	124														
80	Carroll	Holiday Isl #2	T21N,R26W,Sec15	1102	1122					967	500	297	8.4	230		2.8	48.0	27.0	0.050	11.0	19.0	
81	Carroll	Holiday Isl #1	T21N,R26W,Sec17	1010	1058	495				924	600	337	8.3	280	260	2.8	58.0	34.0	0.050	11.0	17.5	
82	Carroll	Holiday Isl	T21N,R26W,Sec22	1500	1184					1354	500											
83	Carroll	Holiday Isl #4	T21N,R26W,Sec27	1520	1880	560		40		1000	502											
84	Crawford	USA #1	T13N,R28W,Sec33	1420	4090	-1510	210	-2075	50													
85	Faulkner	ARGO Edgmon #1	T07N,R12W,Sec06	486	12160	-8314	330	-9384														
86	Franklin	ARKLA Harding #1	T09N,R27W,Sec09	472	7898	-6528		-7068														
87	Franklin	Tnco Conaster 1	T10N,R27W,Sec17	454	8270	-6426		-7031														
88	Fulton	Cty of Salem	T20N,R08W,Sec27	660	1282	255				632		240	7.3	240	245	1.4	50.0	28.0	0.020	1.8	17.0	
89	Fulton	Cty of Viola1	T20N,R09W,Sec18	800	1600	-80	210	-750	40	674	140	298	7.3	220	272	1.5	45.0	26.0	0.200	1.6	16.0	
90	Fulton	Mammoth Spgs	T21N,R05W,Sec07	575	1340	415		-305														
91	Fulton	Mammoth Spgs	T21N,R05W,Sec08	580	363					506	445											
92	Fulton	Rvr Side Resort	T21N,R05W,Sec33	490	370	240					110											
93	Izard	Cty of Melborne1	T16N,R09W,Sec01	580	1446	-750																
94	Izard	Cty of Melborne2	T16N,R09W,Sec01	754	1807	-776	240															
95	Izard	Clco Rck #6	T17N,R11W,Sec12	776	2134	-684	180	-1024	25													
96	Izard	Cty of Clco Rck	T17N,R11W,Sec13	541	1729					464						2.3						17.5
97	Izard	Horsebend #4	T18N,R07W,Sec08	840	1406	-70	240															
98	Izard	Franklin2	T18N,R07W,Sec31	670	1300	-360	250				300	252	8.4	270		2.0	55.0	33.0	0.340	13.0		
99	Izard	Oxford #1	T18N,R09W,Sec23	800	1070	-180																
100	Lawrence	Thatcher #1	T16N,R03W,Sec15	300	2640	-1209		-2081														
101	Lawrence	Lawrence Co. RWA	T17N,R03W,Sec13	500	1010	-500					350											
102	Lawrence	Guthrie #1	T17N,R03W,Sec22	380	1380	-520	230															
103	Lawrence	Smithville #1	T17N,R03W,Sec24	450	1400	-690																
104	Lawrence	Imboden #1	T18N,R02W,Sec21	390	1740	-510	250	-960	10		200											
105	Lawrence	Imboden #2	T18N,R02W,Sec23	340	1260	-540	260															
106	Madison	Huntsville #2	T16N,R26W,Sec03	1670	3288	0		-355		1160	41	299	7.8	51	237	6.9	11.0	5.7	0.140	0.4	20.5	
107	Madison	Huntsville #1	T16N,R26W,Sec03	1670	3290	30		-325	55	1160	41	389	7.2	130		110.0	30.0	15.0	0.002	49.0		

Well	County	Well Name	Location	Elev.	TD	Rbx Top	Rbx Thick	Gtr Top	Gtr Thick	WL	Yield	TDS	pH	Hardness	Alkal	Cl	Ca	Mg	Fe	SO4	Temp (C)	
108	Madison	Bank #1	T16N,R27W,Sec06	1545	2515	59	250	-505	60													
109	Madison	Max B Hoeme	T17N,R25W,Sec21	1900	2290					1221	200											
110	Madison	Gal BG Noark	T17N,R26W,Sec15	1400	1120					1207	100	235	8.1	140		7.7	31.0	14.0			14.0	
111	Madison	#1 Swift-Madison	T17N,R26W,Sec27	1436	2350	76	170	-364	50													
112	Madison	Ozark Science Ct	T18N,R26W,Sec02	1374	1110	544	150				90											
113	Madison	Madison Co. WA	T18N,R26W,Sec35	1558	1970	130				1196	219	177	8.1	160		3.0	34.0	19.0	0.210		20.0	
114	Marion	Buffalo Point	T17N,R15W,Sec34	545	1300	-565				425	40											
115	Marion	Flippin Water #2	T19N,R15W,Sec20	650	1150	-240				530	40											
116	Marion	Cty of Flippin	T19N,R15W,Sec20	630	900					544	30	281	7.8	290	337	2.5	58.0	35.0	0.050		18.0	
117	Marion	Cty of Summit	T19N,R16W,Sec32	950	1524					561	61	275	7.7	280		9.2	58.0	33.0	0.050		7.0	
118	Marion	Cty of Yellville	T19N,R16W,Sec33	840	753					515	60	344	7.4	340		9.8	70.0	39.0	0.090		23.0	
119	Marion	Town of Pyatt #1	T19N,R18W,Sec36	758	1392	-187				732	236											
120	Marion	Corps Engs	T20N,R15W,Sec06	725	506					690	100	420	7.7		354	10.0	79.0		0.001			
121	Marion	Brown Oil	T20N,R15W,Sec08	775	775					649	75											
122	Marion	Buck Creek Park	T21N,R17W,Sec09	720	525	300				610	92											
123	Marion	Corps Engs	T21N,R17W,Sec22	755	525					687	100											
124	Newton	Nail-Swain #1	T13N,R21W,Sec05	2060	3900	-1060	200	-1720	100		125	685	7.4	112		120.0			0.030		131.0	
125	Newton	Piercetown	T15N,R20W,Sec11	980	2520	-580	250	-1430	60		200											
126	Newton	Mockingbird Hill	T15N,R21W,Sec13	2135	3142	-705				839	62	350	8.3	140		41.0	32.0	14.0	0.110		37.0	
127	Newton	Hasty #1	T16N,R20W,Sec25	1230	2590	-510	200	-1220	50		80											
128	Newton	Mt Sherman #2	T16N,R22W,Sec26	2105	3080	-805					58											
129	Newton	Mt Sherman #1	T16N,R22W,Sec27	2240	3288	-785					50											
130	Newton	Marble Falls #1	T17N,R20W,Sec21	1344	2580	-356		-1006	100	996	222	261	7.5	230		12.0	54.0	23.0	0.050		17.0	
131	Newton	Cty of Compton 1	T17N,R22W,Sec26	2110	3215	-475		-1010		1135	52	121	7.3	97	116	3.6	24.0	9.1	0.070		28.0	24.0
132	Randolph	Ravenden Sprgs	T19N,R02W,Sec06	480	875	-230																
133	Randolph	LB Adams #1	T20N,R01E,Sec13	400	925	-5	205															
134	Randolph	Cty of Maynard 1	T20N,R02E,Sec06	485	900	-55				395												
135	Searcy	Cty of Leslie #1	T14N,R15W,Sec15	1060	3535	-1400	215	-2190	30	708	40	358	7.7	180	320	17.0	44.0	18.0	0.190		17.0	15.0
136	Searcy	Marshall (S Mtn)	T14N,R16W,Sec08	880	3700	-1270	220	-1850	80		125											

Well	County	Well Name	Location	Elev.	TD	Rbx Top	Rbx Thick	Gtr Top	Gtr Thick	W L	Yield	TDS	pH	Hard-ness	Alkal	Cl	Ca	Mg	Fe	SO4	Temp (C)	
137	Searcy	Morning Star 2	T15N,R15W,Sec02	1110	2690	-850	280	-1510	20	881	120											
138	Searcy	Marshall #3	T15N,R16W,Sec25	1045	2415	-1045	210			840	55	265	7.5	230	239	5.5	67.0	16.0	0.010	16.0	15.5	
139	Searcy	Morning Star 1	T16N,R15W,Sec33	1240	2900	-800		-1490														
140	Searcy	Tyler Bend #1	T16N,R17W,Sec35	768	2200	-1022	210															
141	Searcy	Wylie Moore 1	T16N,R18W,Sec02	1150	3123	-333	161	-980	49													
142	Searcy	Pndt-Glbt WA 2	T16N,R18W,Sec08	1250	1910	-390	210															
143	Sebastian	Shell Oil #1	T07N,R32W,Sec36	505	10924	-9290		-9860														
144	Sharp	Calamine-Grauge1	T16N,R04W,Sec23	395	1900	-1105	190				190											
145	Sharp	Grange-Calamine2	T16N,R04W,Sec29	460	1850	-1080	150				115											
146	Sharp	Sidney	T16N,R06W,Sec34	742	2700	-778	250	-1488	30		150											
147	Sharp	Grange-Calamine3	T17N,R05W,Sec25	488	1590	-542	210															
148	Sharp	Baxter Lab. #1	T18N,R06W,Sec02	600	1250	-100				578	500											
149	Sharp	Ash Flat	T18N,R06W,Sec10	655	1525					527	180	282		280		3.0	52.0	30.0		16.0	17.0	
150	Sharp	Cty of Ash Flat1	T18N,R06W,Sec11	650	1545	-100	220	-870	25	502	180											
151	Sharp	Ozark Acres3	T19N,R04W,Sec15	590	611					574	120	287	7.9	260		5.0	46.0	32.0	0.010	11.0	16.5	
152	Sharp	Ozark Acres	T19N,R04W,Sec22	473	940					407	125											
153	Sharp	Ozark Acres #1	T19N,R04W,Sec23	680	940	160				614	125											
154	Sharp	Hardy #2	T19N,R05W,Sec02	570	1590	120	160	-730	30													
155	Sharp	Highland #2	T19N,R05W,Sec04	682	1425	102	200	-648	40		50											
156	Sharp	Hardy #1	T19N,R05W,Sec11	375	1180	85	160	-745	50		300											
157	Sharp	Highland #2A	T19N,R05W,Sec15	623	1370	143	170	-667	50		150											
158	Stone	Twn of Fifty-six	T15N,R12W,Sec02	985	3426	-1295		-2285	30	586	40	148	8.4	130		2.5	28.0	15.0	0.110	22.0	20.0	
159	Stone	Big Flat #2	T15N,R13W,Sec06	1130	3105	-1130		-1860	50		150											
160	Washington	Jim Rush	T13N,R33W,Sec34	1700	2439					940	40											
161	Washington	JW Grissom 1	T15N,R31W,Sec17	1140	2097	-342	143	-860	25	990	43											
162	Washington	Holcomb D	T15N,R31W,Sec17	1200	2097					1073	60	93	8.2	93		5.8	19.0	11.0	0.020	8.0		
163	Washington	Ramsey Berlin	T15N,R33W,Sec08	1175	1563					1060	20											
164	Washington	Walker #1	T16N,R31W,Sec06	1220	1525	20	132															
165	Washington	Wedington	T16N,R32W,Sec09	1135	1815	-70	210	-595	40	1034	26	381	7.6	118		114.0	32.8	8.5	0.060	7.0		

Well	County	Well Name	Location	Elev.	TD	Rbx Top	Rbx Thick	Gtr Top	Gtr Thick	W L	Yield	TDS	pH	Hardness	Alkal	Cl	Ca	Mg	Fe	SO4	Temp (C)	
166	Washington	Jim Cash	T16N,R33W,Sec20	1125	1402					955	30											
167	Washington	Tyson Foods	T16N,R33W,Sec36	1510	1806					1071	20											
168	Washington	RW White River	T17N,R29W,Sec04	1420	2080					1005	300	326	7.5	94	204	61.0	22.0	9.6	<.003	8.3	21.0	
169	Washington	RDC White River	T17N,R29W,Sec09	1481	2071	64	204	-528		1066	400					34.0					20.5	
170	Washington	St. Joseph Crch	T17N,R31W,Sec01	1320	1416					1065		289	7.7	84		44.0	17.0	10.0			6.0	
171	Washington	Cty of Tontitown	T17N,R31W,Sec06	1320	1416	104	134															
172	Washington	Casey Raymond	T17N,R31W,Sec07	1000	1505					971		257	8.2	110		49.0	25.0	11.0			11.0	
173	Washington	Bobby Pianalto	T17N,R31W,Sec14	1288	1591					1047	200											
174	Yell	OXY USA Dnvl 4	T05N,R22W,Sec33	848	20661	-19602																

